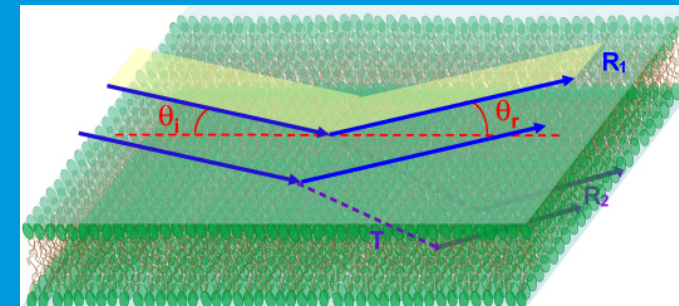
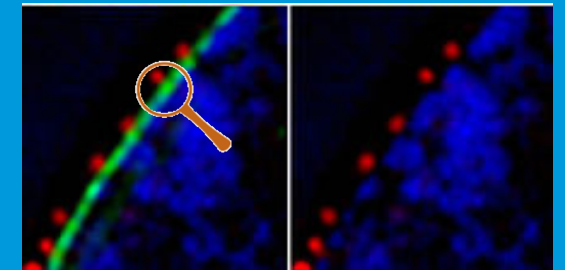
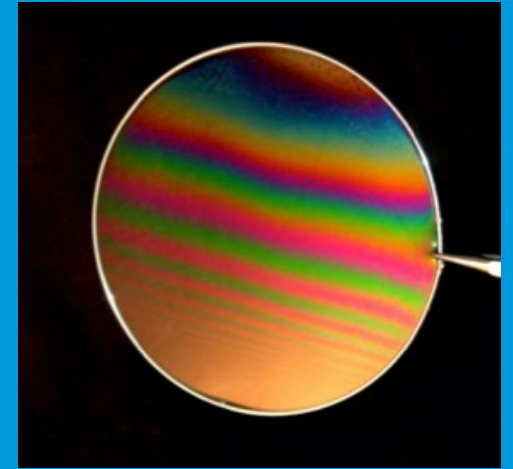


Membrane deuteration and internal contrast

Examples of applications in NR
(incl. lipolytic enzymes!)

HANNA WACKLIN-KNECHT, ESS DEMAX PLATFORM

2021-06-17





We will look today at:

- 1 Contrast variation in membrane samples – why and how?
- 2 Deuteration of membrane lipids
- 3 Data analysis considerations
- 4 Example 1: Phospholipase A_2 enzymes
- 5 Example 2: Lipid asymmetry
- 6 Example 3: Sterol extraction by Amphotericin B

2

Contrast variation – why and how?

$$\text{Contrast} = (\rho_{\text{object}} - \rho_{\text{environment}})^2$$



When the monster came,
Lola remained undetected.

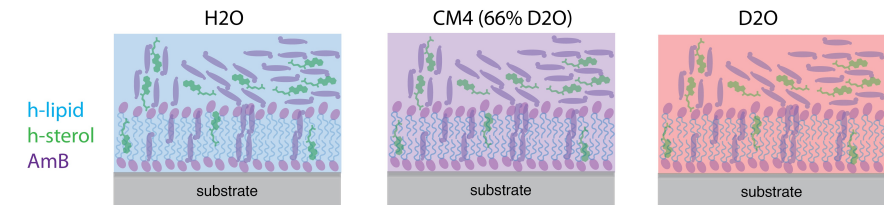
Harold, of course, was immediately devoured.

1 Contrast variation in membranes

Solvent contrast, membrane deuteration or magnetic contrast?

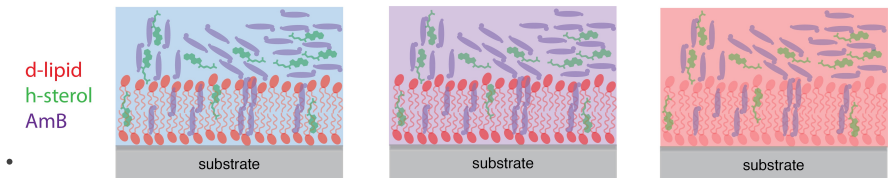
A) Solvent contrast variation:

Same sample measured in solvents with a different degree of deuteration (0-100% D₂O) -> simultaneous fit to structure in principle solves phase problem and **allows SLD profile to be determined**. -> *overall membrane structures, solvent content*



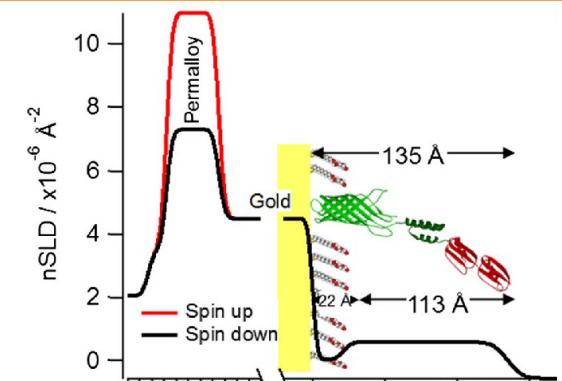
B) Membrane deuteration:

Multicomponent membranes -> **selective deuteration** allows composition and component distribution to be determined (e.g. two different lipids, or a protein and lipids) -> *combine with A)*



C) Magnetic contrast and polarised neutron reflection:

Permalloy layer has different SLD for up/down spins -> **two different measurement contrasts** – longer measurement times, but no contrast matching possible -> *combine with A) and or B).*





2 Lipid deuteration

Biological, enzyme catalysed and synthetic chemistry

Organic synthesis of lipids

Pure molecular species in mg-g quantities, partial or per-deuteration, contrast matching -> complex synthesis routes

Enzyme-catalysed synthesis of lipids

Lipases and phospholipases shorten lipid synthesis routes due to regio- and enantiospecificity -> immobilisation increases yield

Biological deuteration of lipids in cell cultures

Microorganisms such as yeast, bacteria, algae can grow in deuterated media -> 10-500mg biological lipid mixtures with native fatty acyl chain distribution in per-deuterated, partially deuterated or contrast matched forms. Separation of components by chromatography.

PC, (PG), tri-, di-,
monoglycerides
Fatty acids: saturated +
C18:1c9

PC, PE

Yeast, E.Coli

Phospholipids: PC, PE, PS, PG, PI,
CL

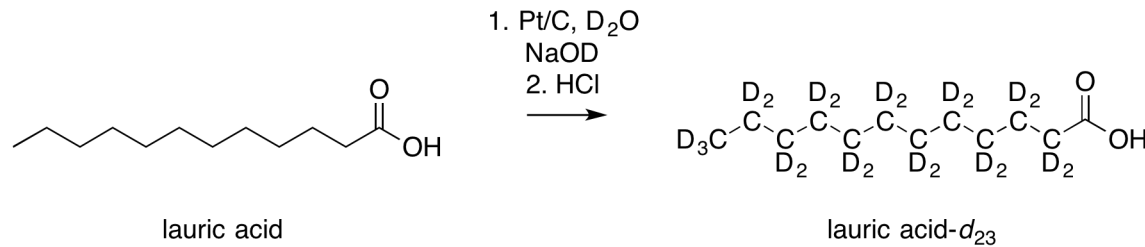
Sterols: Ergosterol, cholesterol
(squalene, lanosterol)

Glycerolipids: TG, DG, MG

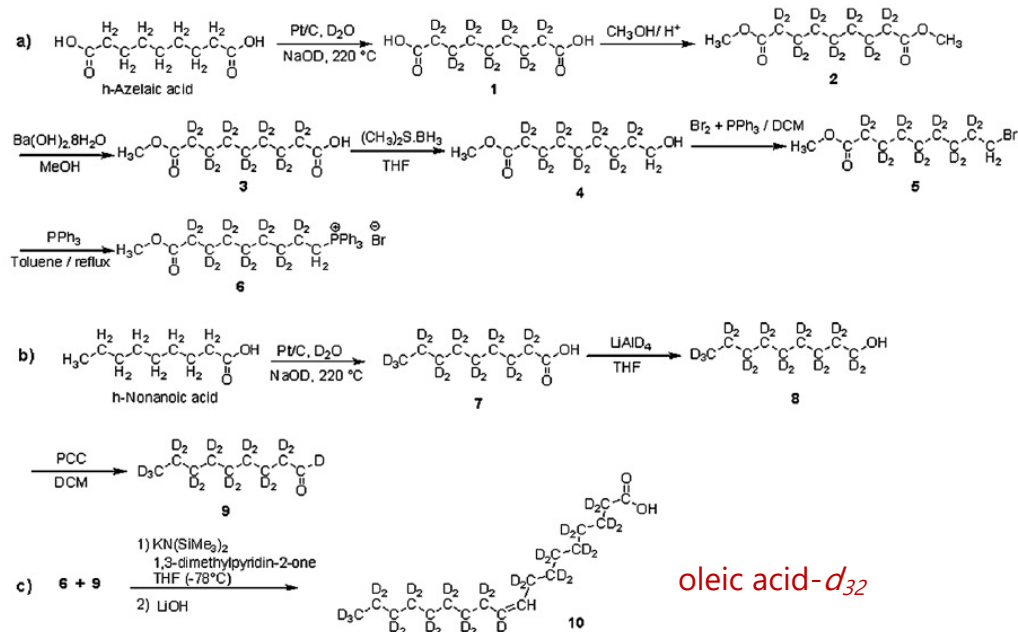
2 Lipid deuteration

Organic synthesis of lipids

Perdeuteration of saturated fatty acids by hydrothermal H/D exchange:



Deuteration of un-saturated fatty acids from saturated precursors:



Special Issue Article

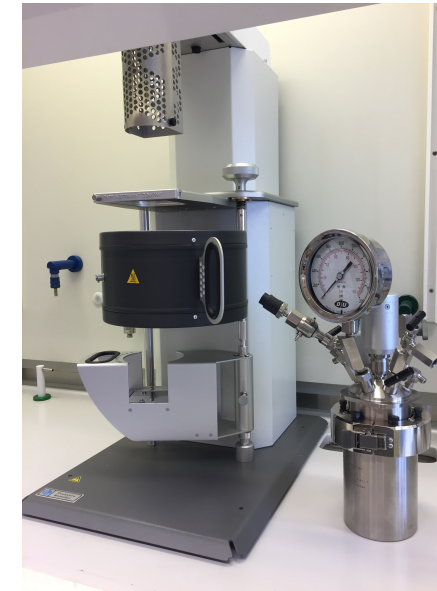
Received 22 May 2013, Revised 12 June 2013, Accepted 12 June 2013, Published online in 16 July 2013 Wiley Online Library
 (wileyonlinelibrary.com) DOI: 10.1002/jlcr.3088

Journal of
 Labelled Compounds and
 Radiopharmaceuticals

Synthesis of deuterated [D₃₂]oleic acid and its phospholipid derivative [D₆₄]dioleoyl-*sn*-glycero-3-phosphocholine^{†‡}

Tamim A. Darwish,^{a*} Emily Luks,^a Greta Moraes,^a Nageshwar R. Yepuri,^a Peter J. Holden,^a and Michael James^{a,b,c}

Biologically deuterated oleic acid-d₃₃ commercially available from Merck since 2019 ~3k€ per gram.



2 Lipid deuteration

Organic synthesis of PC lipids

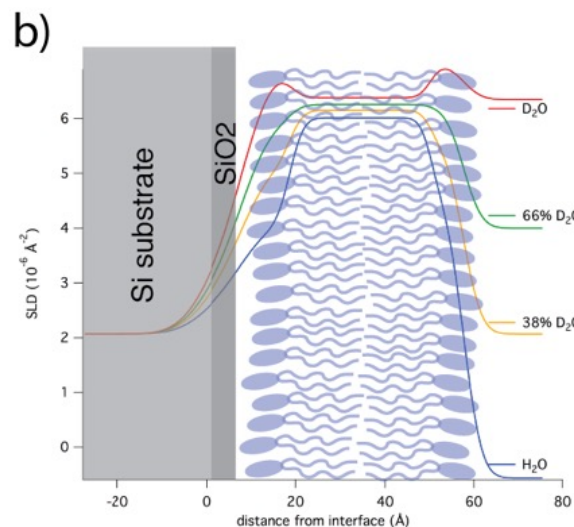
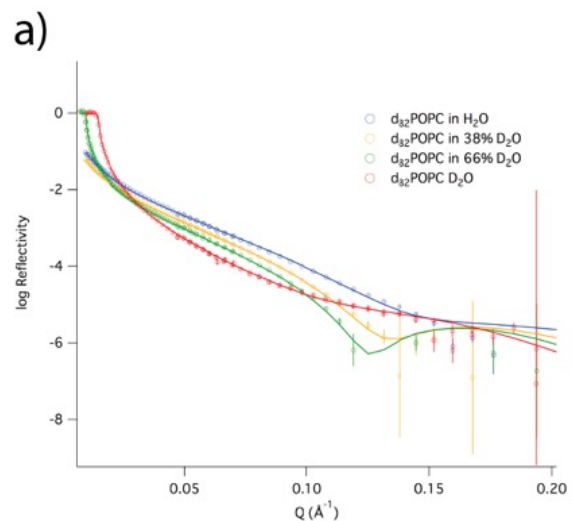
Article
The Antifungal Mechanism of Amphotericin B Elucidated in Ergosterol and Cholesterol-Containing Membranes Using Neutron Reflectometry

Robin Delhom^{1,2,3}, Andrew Nelson⁴, Valerie Laux¹, Michael Haertlein¹, Wolfgang Knecht^{3,5}, Giovanna Fragneto¹ and Hanna P. Wacklin-Knecht^{2,6,*}

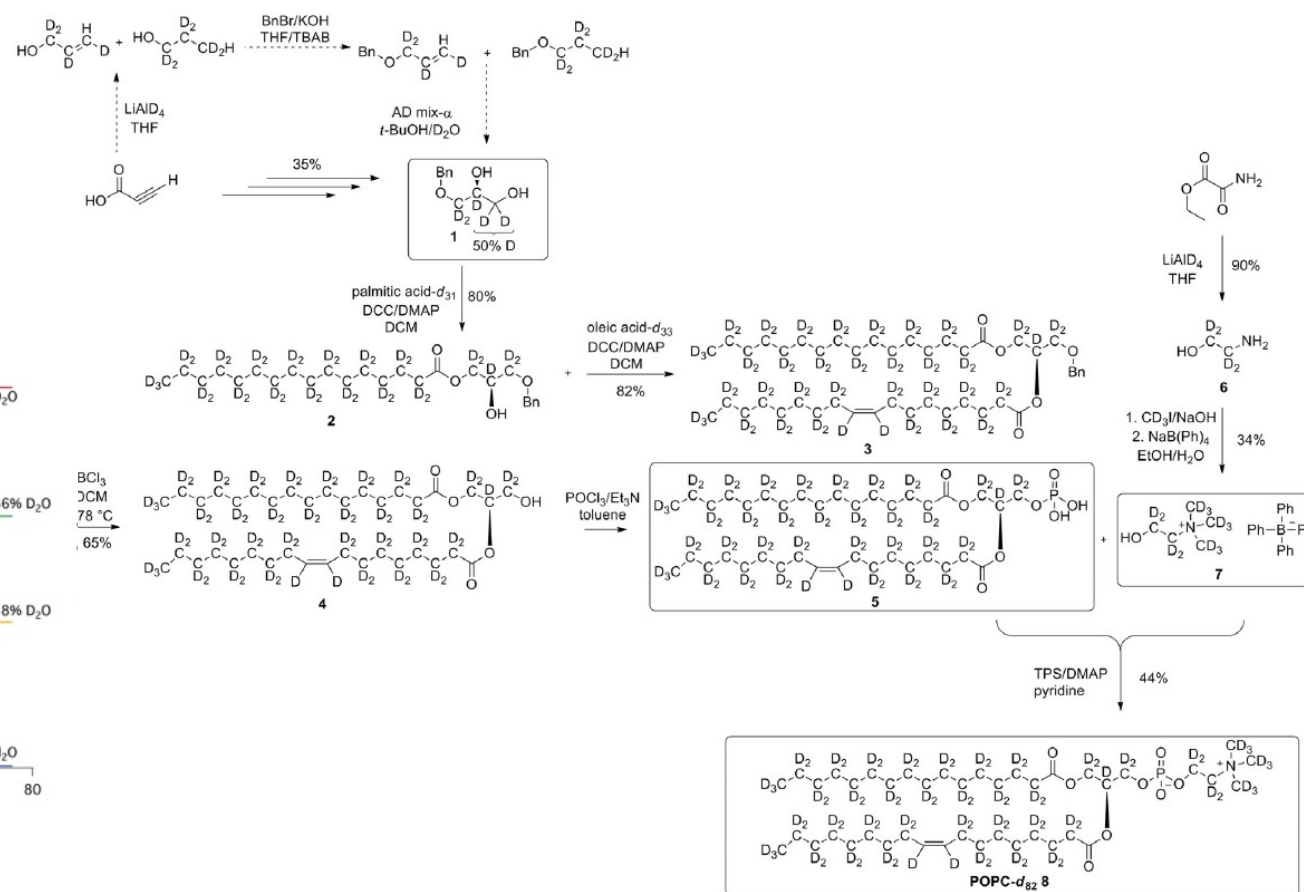
Nanomaterials 2020, 10, 2439 [doi:10.3390/nano10122439](https://doi.org/10.3390/nano10122439)

Synthesis of Perdeuterated 1-Palmitoyl-2-oleoyl-*sn*-glycero-3-phosphocholine ([D₈₂]POPC) and Characterisation of Its Lipid Bilayer Membrane Structure by Neutron Reflectometry

Nageshwar R. Yepuri,^[a] Tamim A. Darwish,^{*,[a]} Anwen M. Krause-Heuer,^[a] Anna E. Leung,^[a] Robin Delhom,^[b, c] Hanna P. Wacklin,^{*,[b, d]} and Peter J. Holden^[a]



ANSTO – NDF/Platypus



2 Lipid deuteration

Enzymes for lipid synthesis

Enzymatic Synthesis

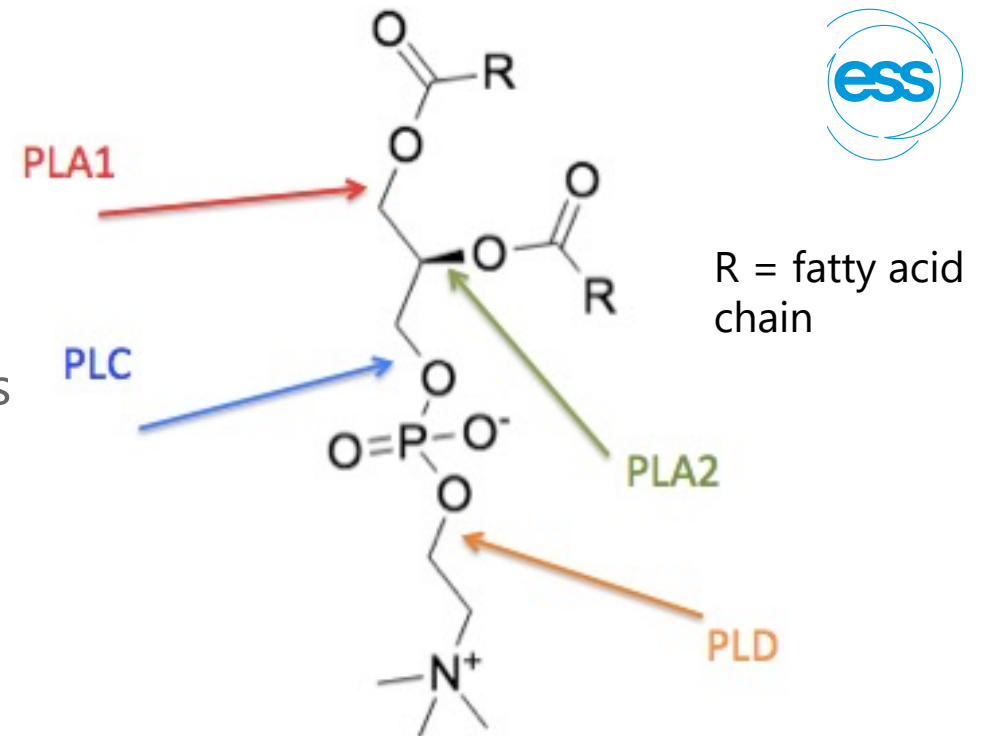
- **CLEAN** and **GREEN** – few by products/no toxic chemicals
- Highly specific – shortens reactions/purifications
- Immobilised enzymes can be reused

Application to lipid deuteration:

- different lipolytic enzymes attack selectively in different positions
- Can be used to swap d-fatty acids h-fatty acids

Commercial lipolytic enzymes available:

- Lipases (1,3 specific), PLA₂

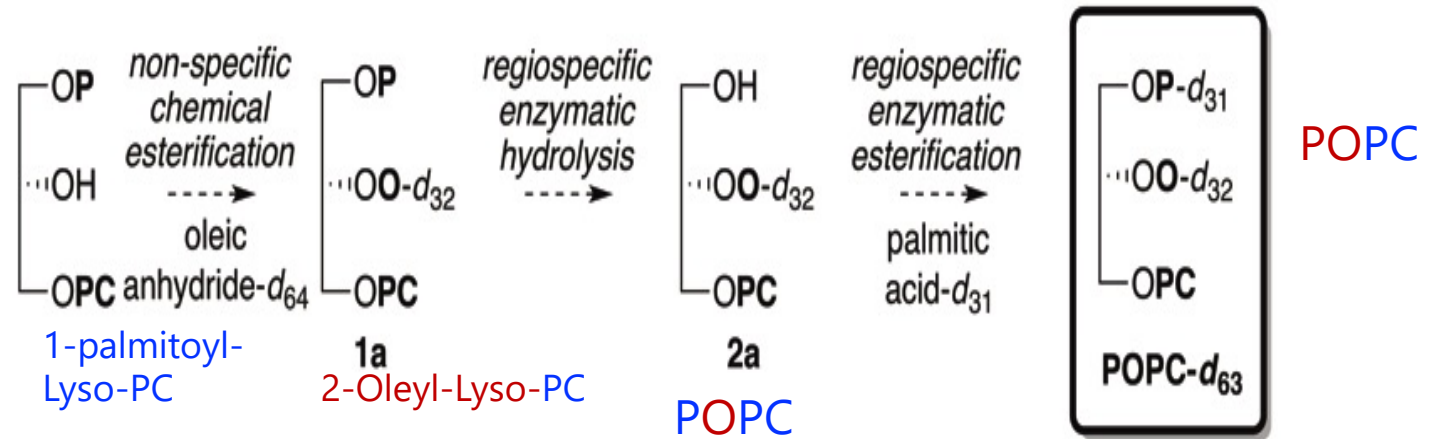


2 Lipid deuteration

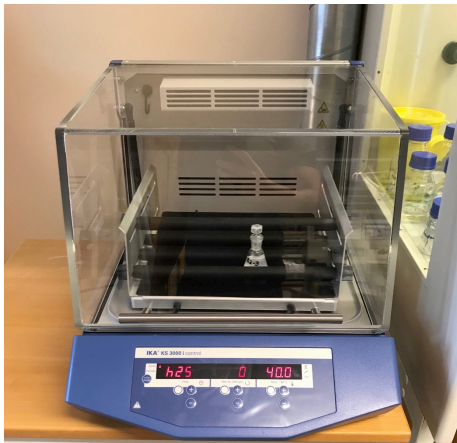
Enzymes for lipid synthesis

Practical considerations:

- Acyl-migration
- Enzyme & water activity
- Solubility & Mixing
- Reaction monitoring
- Enzyme immobilisation



P = palmitoyl; O = oleoyl; PC = phosphocholine. Chirality is not shown but the starting lipid is enantiopure and conservation of enantiopurity is expected throughout the synthesis.



<http://pubs.acs.org/journal/acsofd>

Article

Enzyme-Assisted Synthesis of High-Purity, Chain-Deuterated 1-Palmitoyl-2-oleoyl-*sn*-glycero-3-phosphocholine

Oliver Bogojevic and Anna E. Leung*

Cite This: <https://dx.doi.org/10.1021/acsomega.0c02823>

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SINE 2020

brightness²

2 Lipid deuteration

Biological lipid deuteration – *P. Pastoris* yeast

- *Pichia Pastoris* yeast (used for protein production) grows well in minimal D₂O-media with glycerol-*d*₈ as carbon source
- Growth either in Shaker flasks or bioreactors yields 100-1000mg per-deuterated lipids
- *P. pastoris* produces a significant amount of C18:2 and C18:3 lipids (Δ 12 and Δ 15 desaturase enzymes)
- Deuterated *P. pastoris* produces the same phospholipid composition but fewer polyunsaturated lipids and less ergosterol
- Difference in PUFA chains reduced at lower growth T in D₂O
- Growth rate/Kinetic isotope effect behind difference.

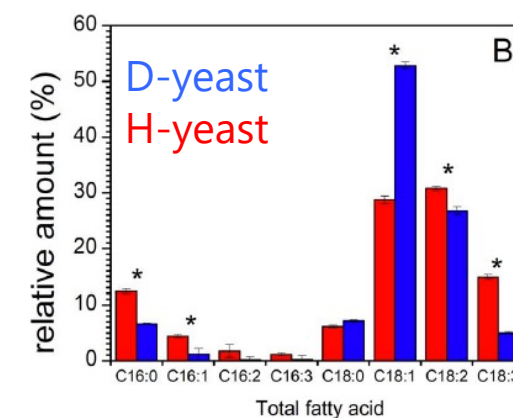
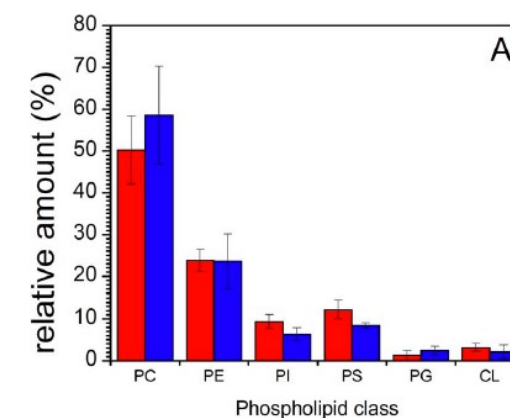
Production and Analysis of Perdeuterated Lipids from *Pichia pastoris* Cells

Alexis de Ghellinck^{1,2}, Hubert Schaller³, Valérie Laux¹, Michael Haertlein¹, Michele Sferrazza², Eric Maréchal⁴, Hanna Wacklin^{5,6}, Juliette Jouhet^{4*}, Giovanna Fragneto¹

¹ Institut Laue-Langevin, Grenoble, France, ² Service des polymères, Université Libre de Bruxelles, Brussels, Belgium, ³ Institut de Biologie Moléculaire des Plantes du CNRS, Strasbourg, France, ⁴ Laboratoire de physiologie cellulaire et végétale, CNRS/CEA/Univ. Grenoble Alpes/INRA, Grenoble, France, ⁵ European Spallation Source ESS AB, Lund, Sweden, ⁶ Chemistry Department, University of Copenhagen, Copenhagen, Denmark

Abstract

Probing molecules using perdeuteration (i.e. deuteration in which all hydrogen atoms are replaced by deuterium) is extremely useful in a wide range of biophysical techniques. In the case of lipids, the synthesis of the biologically relevant unsaturated perdeuterated lipids is challenging and not usually pursued. In this work, perdeuterated phospholipids and sterols from the yeast *Pichia pastoris* grown in deuterated medium are extracted and analyzed as derivatives by gas chromatography and mass spectrometry respectively. When yeast cells are grown in a deuterated environment, the phospholipid homeostasis is maintained but the fatty acid unsaturation level is modified while the ergosterol synthesis is not affected by the deuterated culture medium. Our results confirm that the production of well defined natural unsaturated perdeuterated lipids is possible and gives also new insights about the process of desaturase enzymes.

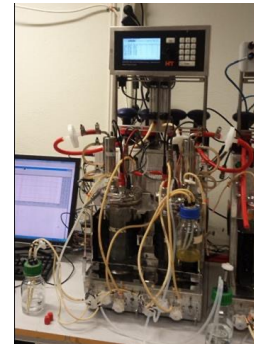
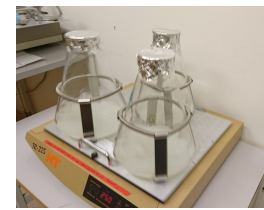


2 Lipid deuteration

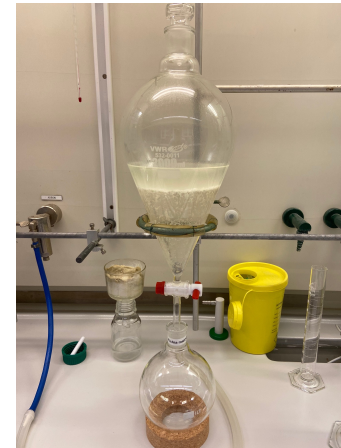
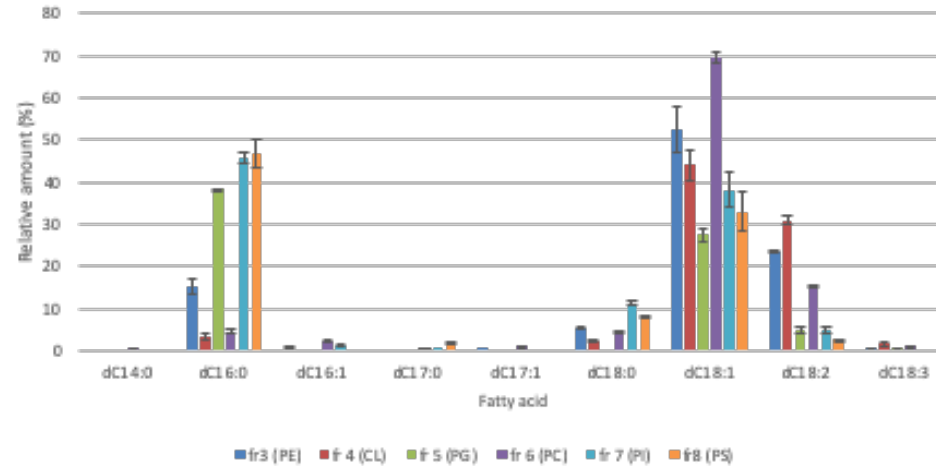
Lipid extraction and purification – yeast

- Phospholipids are not overexpressed by cells
- > Large scale growth/extraction/purification
- Main cost **glycerol- d_8**
- Non-polar/phospholipid lipid separation
- **Growth conditions affect lipid composition**
- > Analysis of composition (TLC, GC, MS)
- PC, CL and ergosterol/cholesterol* available purified from ESS and ILL.

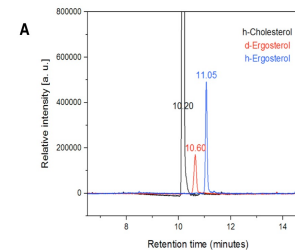
Shaker flasks
 Fermentor/bioreactors
 -> 100mg - 1 g of lipids



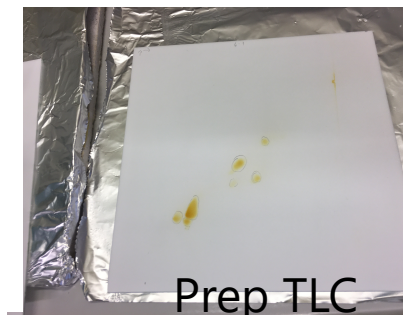
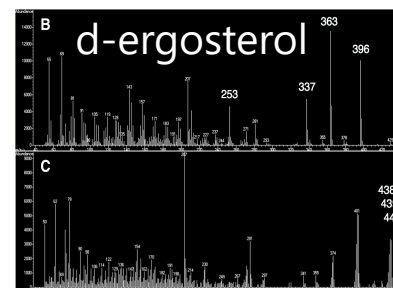
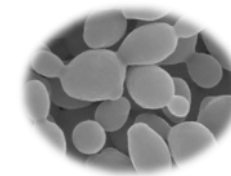
Composition of fatty acids deuterated *Pichia* Phospholipids



Extraction/separation

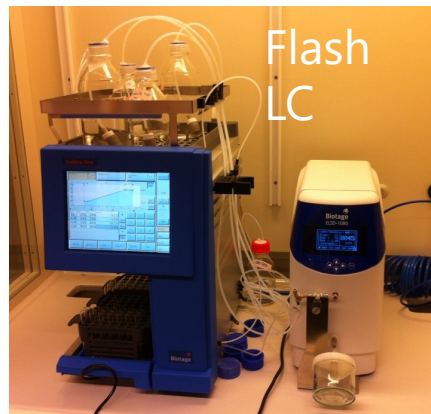
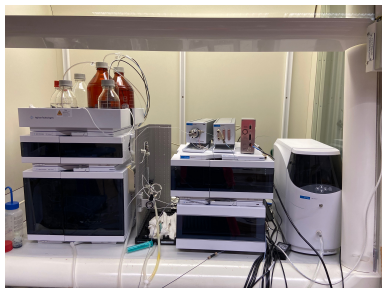


Pichia pastoris



Prep TLC

Prep HPLC



Flash LC



GC

2 Lipid deuteration

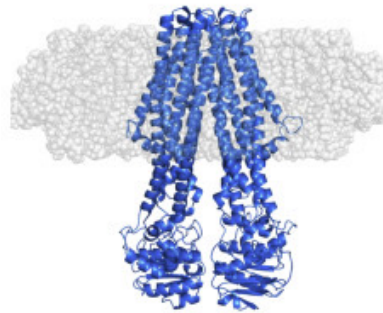
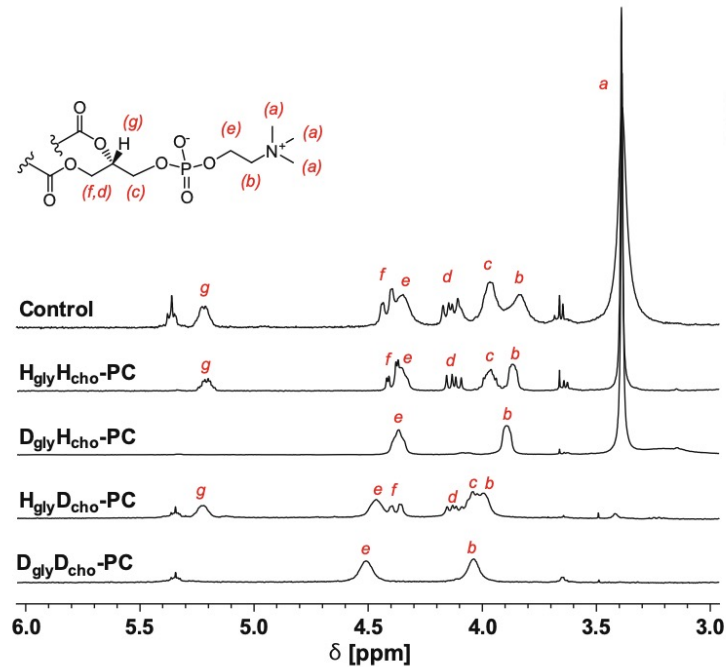
Biological lipid deuteration – *E.Coli*

Appl Microbiol Biotechnol (2015) 99:241–254
DOI 10.1007/s00253-014-6082-z

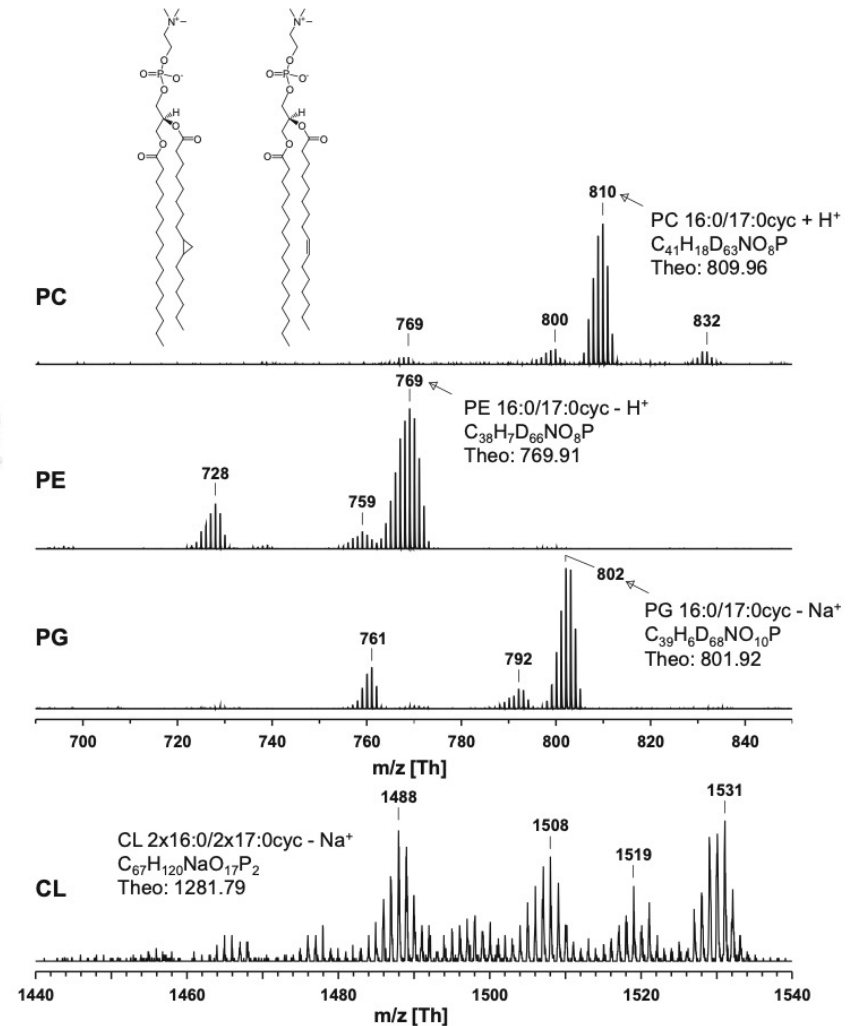
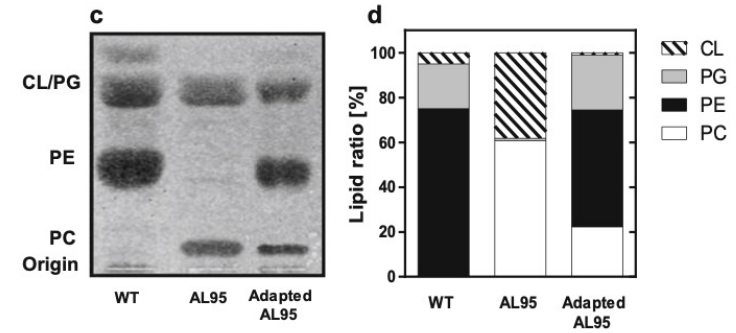
BIOTECHNOLOGICAL PRODUCTS AND PROCESS ENGINEERING

Biosynthetic preparation of selectively deuterated phosphatidylcholine in genetically modified *Escherichia coli*

Selma Maric · Mikkel B. Thygesen · Jürgen Schiller · Magdalena Marek ·
Martine Moulin · Michael Haertlein · V. Trevor Forsyth · Mikhail Bogdanov ·
William Dowhan · Lise Arleth · Thomas Günther Pomorski



I. Josts, J. Nitsche, S. Maric, et al.,
Structure 26(8)
(2018) 1072-1079.e4.

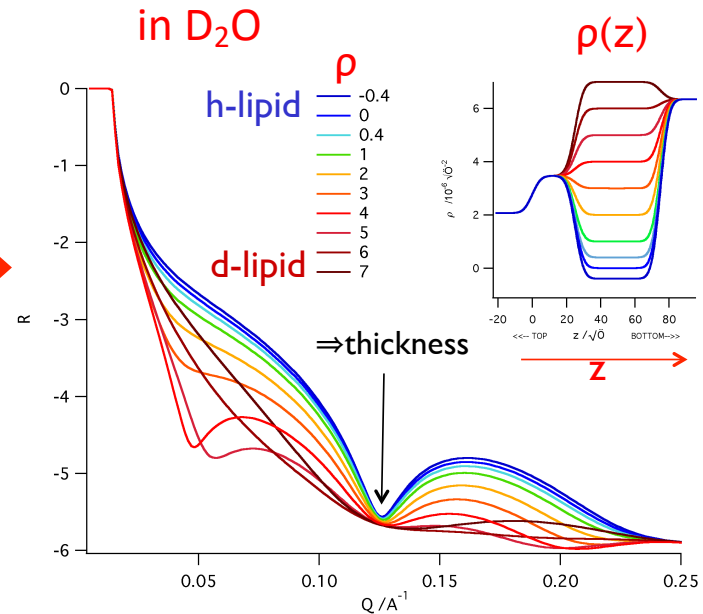
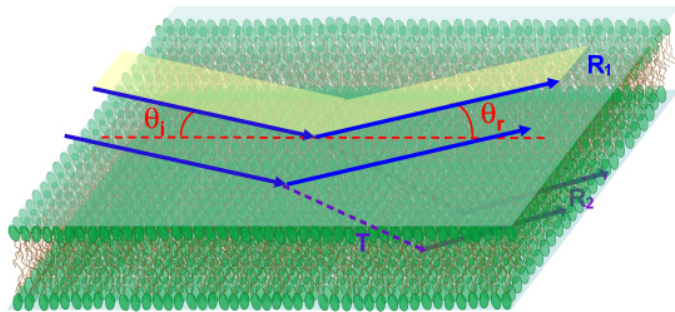


Deuteration in Neutron reflectometry



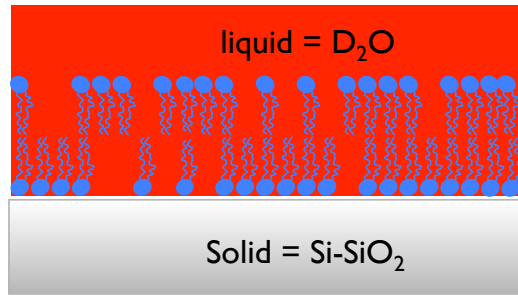
Selective deuteration of lipids to create contrast between components

Composition sensitivity $\sim 3\text{-}5\text{ Vol}\%$

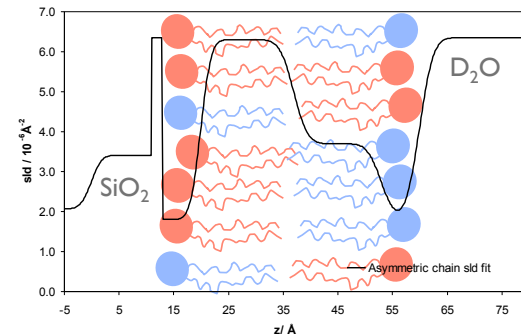


1-2Å resolution

z



Model structure of lipid membrane
= layered model of structure & composition

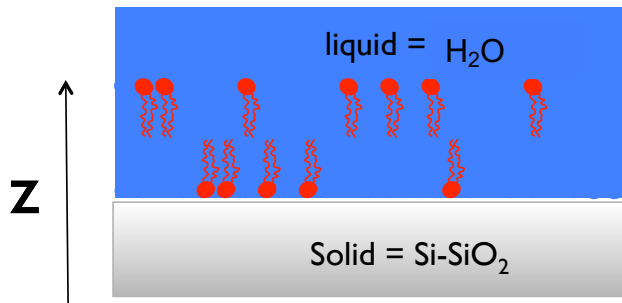
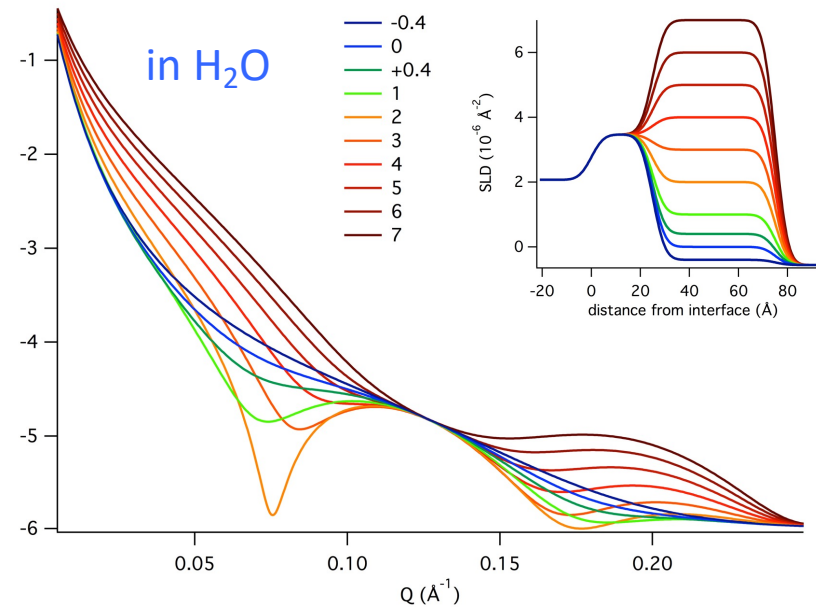
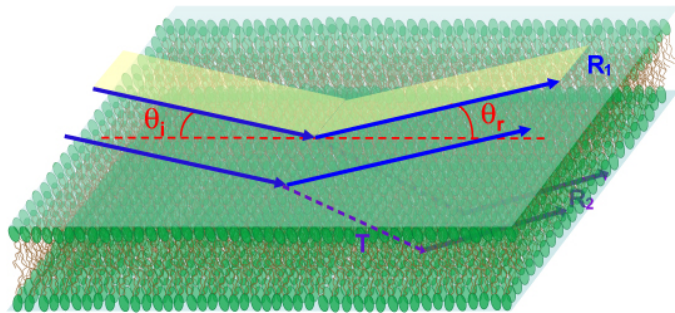


Deuteration in Neutron reflectometry

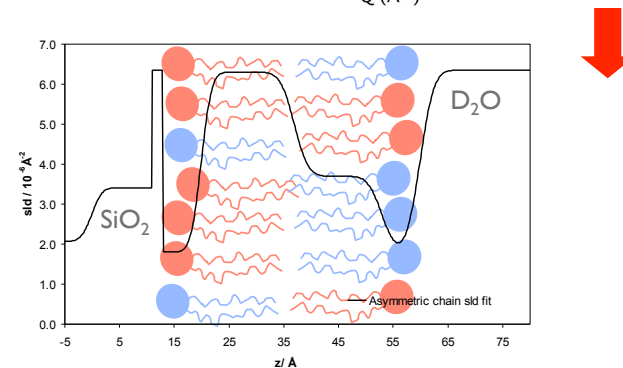
Solvent contrast variation to highlight/contrast match out components



Contrast determines sensitivity



Model structure of lipid membrane
= layered model of structure & composition



Scattering length density profile $\rho(z)$
= neutron refractive index profile

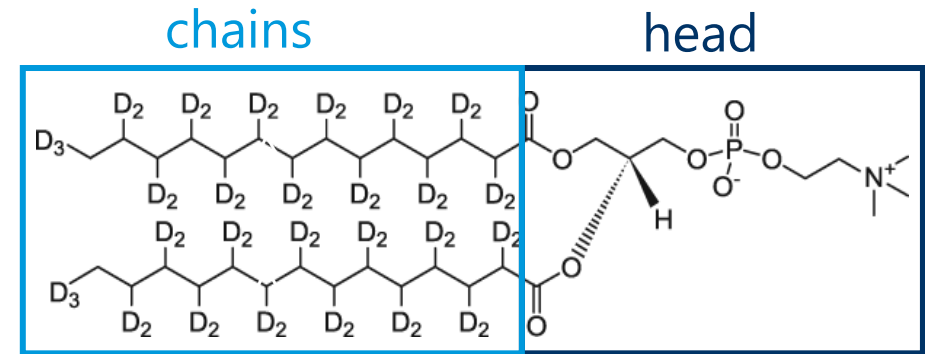
3 Data analysis considerations

What do you need to know (A) and do (B) for NR analysis?

A) The expected SLD of your molecule(s)

- Degree of deuteration
- Chemical composition
- Molecular structure
- Molecular volume or density
- Exchange of protons with solvent

$$SLD = \frac{\text{sum of } b_{coh}}{V_m}$$



Armen, R.S., O.D. Uitto, and S.E. Feller, *Phospholipid component volumes: Determination and application to bilayer structure calculations*. Biophysical Journal, 1998. **75** 734-744.

B) Stoichiometry/geometry and interpretation

- Constant lipid area per molecule (heads/tails)
- Space-filling/geometry
- Analysis of experimental and fitting uncertainties

Group	$V(\text{DPPC})/\text{\AA}^3$	$V(\text{DOPC})/\text{\AA}^3$	$V(\text{POPC})/\text{\AA}^3$
CH ₃	53.57±0.91	52.79±0.4	50.41±0.77
CH ₂	27.93±0.1	28.13±0.09	28.24±0.17
C=C		45.91±0.69	42.1±2.31
COO Carbonyl	43.58±1.05	37.4±1.28	38.43±1.47
C ₃ H ₅ Glycerol backbone	65.47±2.11	81.62±2.96	72.48±2.28
PO ₄ Phosphate	66.36±2.95	51.05±6.31	52.12±2.23
C ₅ H ₁₃ N Choline	107.62±2.49	129.68±5.25	120.68±2.98
H ₂ O	29.5±0.5	29.5±0.5	29.5±0.5
Total	1215.79±14.27	1322.19±20.52	1255.78±17.12
Experimental [7-10]	1232	1295	1267

[7] Costigan, S.C., P.J. Booth, and R.H. Templer, BBA-Biomembranes, 2000. **1468** 41-54.

[8] Wiener, M.C. and S.H. White, Biophysical Journal, 1992. **61** 434-447.

[9] Small, D.M., *The Physics and Chemistry of Lipids*. 1986, Plenum Press.

[10] Nagle, J.F. and M.C. Wiener, BBA, 1988. **942** 1-10

3 Data analysis considerations

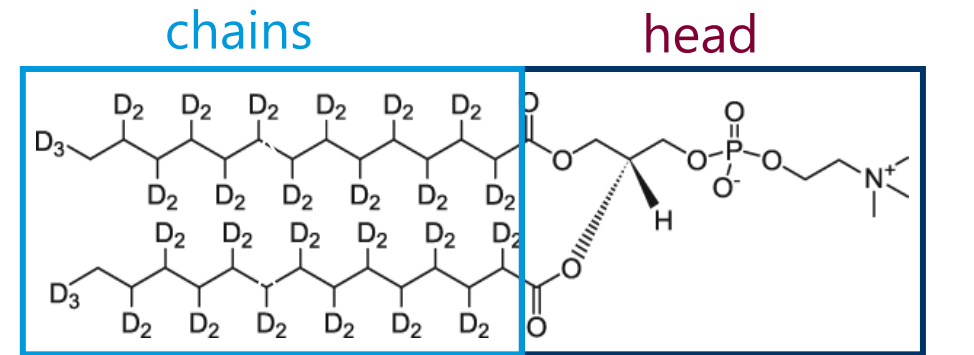
What do you need to know (A) and do (B) for NR analysis?

A) The expected SLD of your molecule(s)

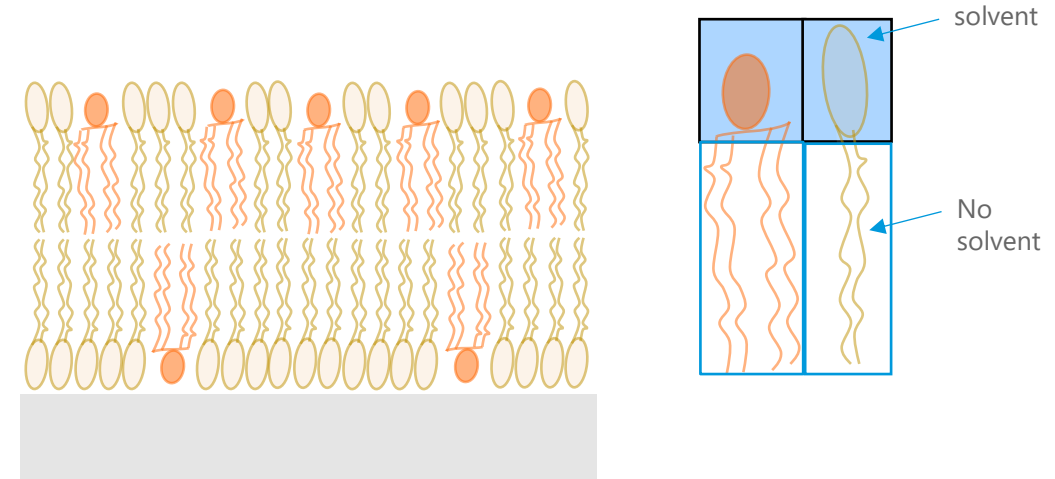
- Degree of deuteration
- Chemical composition/purity
- Chemical formula
- Molecular volume or density
- Exchange of protons with solvent

B) Stoichiometry/geometry and interpretation

- Constant lipid area per molecule (heads/tails)
- Space-filling/geometry
- Analysis of experimental and fitting uncertainties!!!



SLD rough thick vol%



3 Data analysis considerations

What do you need to know (A) and do (B) for NR analysis?

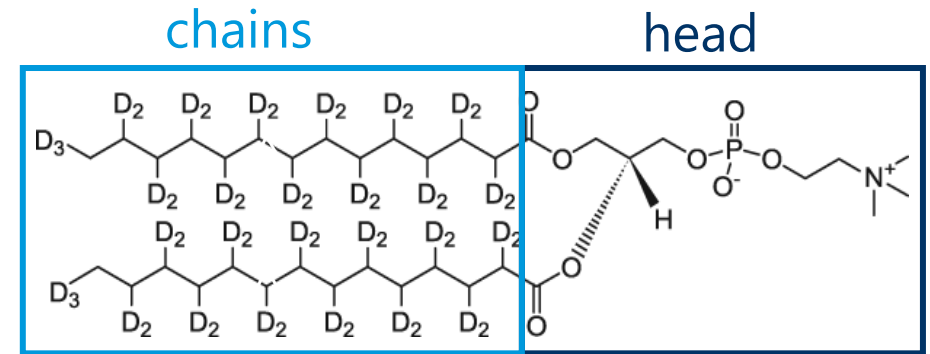
A) The expected SLD of your molecule(s)

- Degree of deuteration
- Chemical composition/purity
- Chemical formula
- Molecular volume or density
- Exchange of protons with solvent

B) Stoichiometry/geometry and interpretation

- Constant lipid area per molecule (heads/tails)
- Space-filling/geometry
- Analysis of experimental and fitting uncertainties!!!

$$A = \frac{V_m}{\phi * t}$$



		SLD	rough	thick	vol%
	layer	$\rho/\text{\AA}^{-2}$	$\sigma/\text{\AA}$	$t/\text{\AA}$	ϕ
D2O	head	1.8 ± 0.3	2 ± 1	6 ± 1	0.83 ± 0.2
	chains	6.8 ± 0.3	2 ± 1	28 ± 1	0.85 ± 0.4
H2O	head	1.8 ± 0.3	2 ± 1	6 ± 1	0.83 ± 0.2
	head	1.8 ± 0.2	2 ± 1	6 ± 2	0.83 ± 0.2
	chains	6.8 ± 0.05	2 ± 1	28 ± 2	0.85 ± 0.05
	head	1.8 ± 0.2	2 ± 1	6 ± 2	0.83 ± 0.2

Global fit to both contrasts: All parameters are the same in both solvents (except solvent SLD and background)

3 Data analysis considerations

What do you need to know (A) and do (B) for NR analysis?

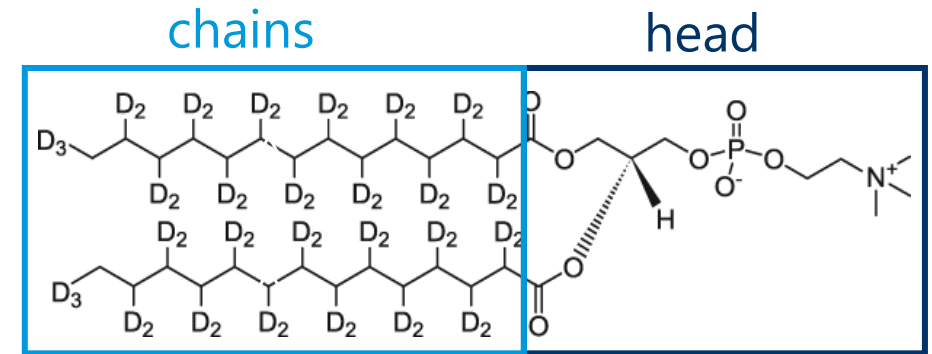
A) The expected SLD of your molecule(s)

- Degree of deuteration
- Chemical composition/purity
- Chemical formula
- Molecular volume or density
- Exchange of protons with solvent

B) Stoichiometry/geometry and interpretation

- Constant area per lipid molecule (heads/tails)
- Space-filling/geometry
- Analysis of experimental and fitting uncertainties
- Best contrast/smallest error for each parameter

$$A = \frac{V_m}{\phi * t}$$



	layer	$V_m / \text{\AA}^3$	$t / \text{\AA}$	ϕ	$A / \text{\AA}^2$
D2O	head	326	6 ± 1	0.83 ± 0.2	66 ± 20
	chains	1555	28 ± 1	0.85 ± 0.4	65 ± 15
H2O	head	326	6 ± 1	0.83 ± 0.2	66 ± 20
	head	326	6 ± 2	0.83 ± 0.2	66 ± 15
	chains	1555	28 ± 2	0.85 ± 0.05	65 ± 5
	head	326	6 ± 2	0.83 ± 0.2	66 ± 15

"Area per molecule": includes any solvent present and should be the same for heads and tails (not necessarily for both leaflets).



3 Data analysis considerations

SLD of yeast phospholipids from analysed lipid composition

Headgroups:

	Chemical formula in H ₂ O	Chemical formula in D ₂ O	V _m (Å ³)	relative amount (%)
h-PC	C ₁₀ O ₈ H ₁₈ PN	C ₁₀ O ₈ H ₁₈ PN	323	53.1
h-PE	C ₇ O ₈ H ₁₂ PN	C ₇ O ₈ H ₉ D ₃ PN	235	22.2
h-PS	C ₈ O ₁₀ H ₁₁ P	C ₇ O ₈ H ₈ D ₃ P	335	13.5
h-PI	C ₁₁ O ₁₃ H ₁₆ P	C ₁₁ O ₁₃ H ₁₁ D ₅ P	372	11.1
d-PC	C ₁₀ O ₈ D ₁₈ PN	C ₁₀ O ₈ D ₁₈ PN	323	48.2
d-PE	C ₇ O ₈ D ₉ H ₃ PN	C ₇ O ₈ D ₁₂ PN	235	20.1
d-PC	C ₇ O ₈ D ₈ H ₃ P	C ₈ O ₁₀ D ₁₁ P	335	17.6
d-PI	C ₁₁ O ₁₃ D ₁₁ H ₅ P	C ₁₁ O ₁₃ D ₁₆ P	372	14.0

Proton exchange of headgroups in different solvents: CM4 = 66% D₂O (SLD 4.0), CmSi = 38% D₂O (SLD 2.07)

	D ₂ O	CM4	CMSi	H ₂ O
h-PC	1.86	1.86	1.86	1.86
h-PE	3.99	3.47	3.11	2.66
h-PI	3.25	3.02	2.85	2.64
h-PS	3.92	3.39	3.00	3.52
Averaged ρ				
	D2O	CM4	CMSi	H2O
	2.82	2.60	2.44	2.25

Chains:

Chains	C16:0	C16:1	C16:2	C16:3	C18:0	C18:1	C18:2	C18:3	
Volume	443,72	436,17	425,82	415,47	499,58	492,43	453,95	471,73	
% in h-lipids	12	5	2	1	7	29	31	16	
%in d-lipids	6	1	0,5	0,5	7	53	27	6	<i>averaged</i>
sld h (x10 ⁻⁶)	-0,386	-0,201	-0,031	0,149	-0,359	-0,212	-0,66	0,096	-0.143
sld d (x10 ⁻⁶)	7,36	6,72	6,57	6,41	6,94	6,77	7,045	6,496	6.61

Higher than POPC due to PUFAs

3 Data analysis considerations

Global fitting of NR data contrasts

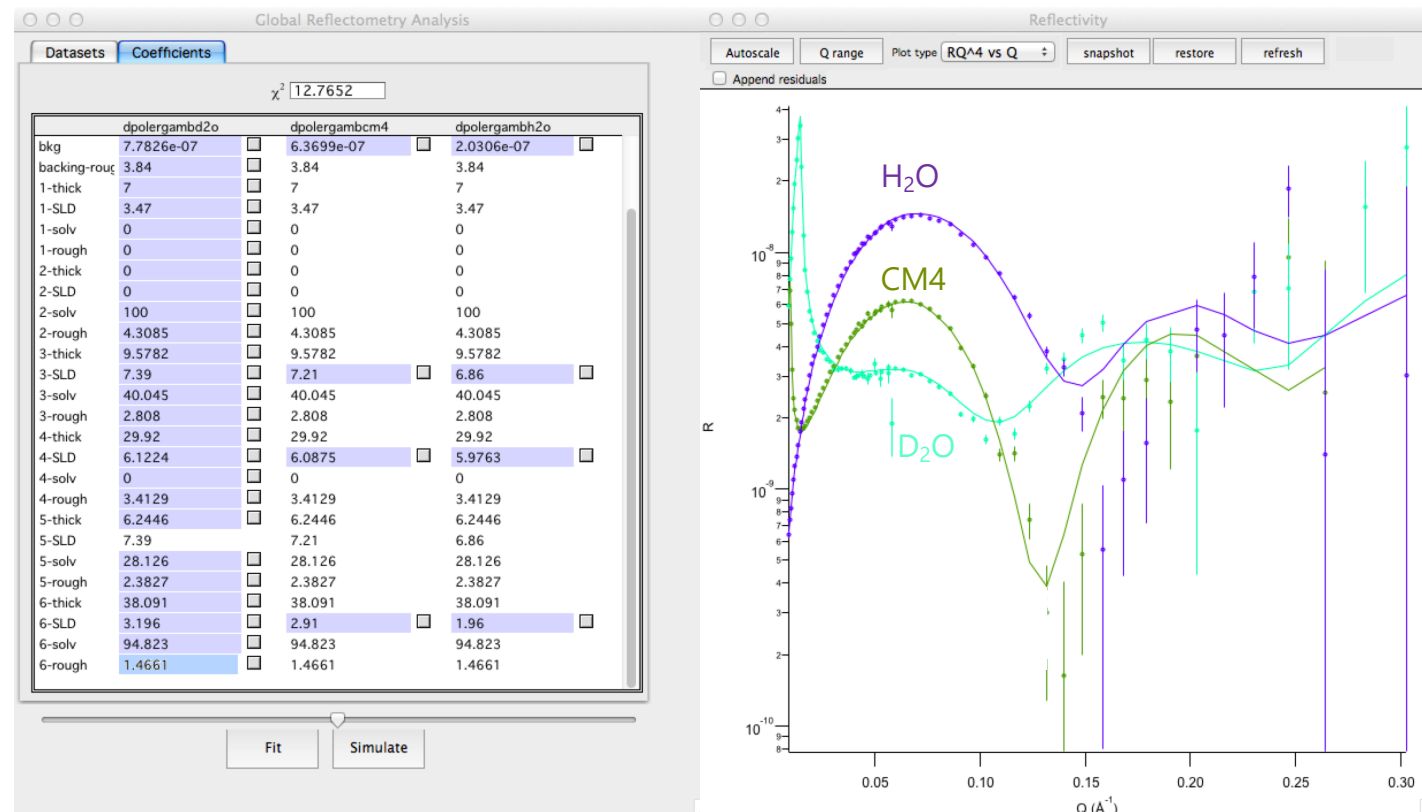
Fits are constrained so that only sld changes due to solvent contrast and other exchangeable protons are allowed.

Supported bilayers: for best resolution and confidence to bilayer structure, each SiO₂ substrate should be characterised clean and used as a reference in the fits (layer 1).

Deuterated yeast phospholipid bilayer model

Layers:

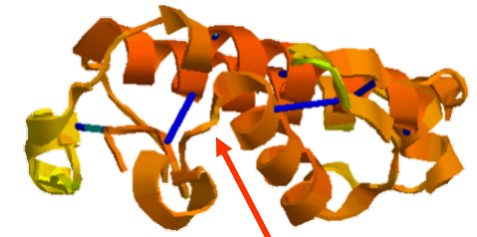
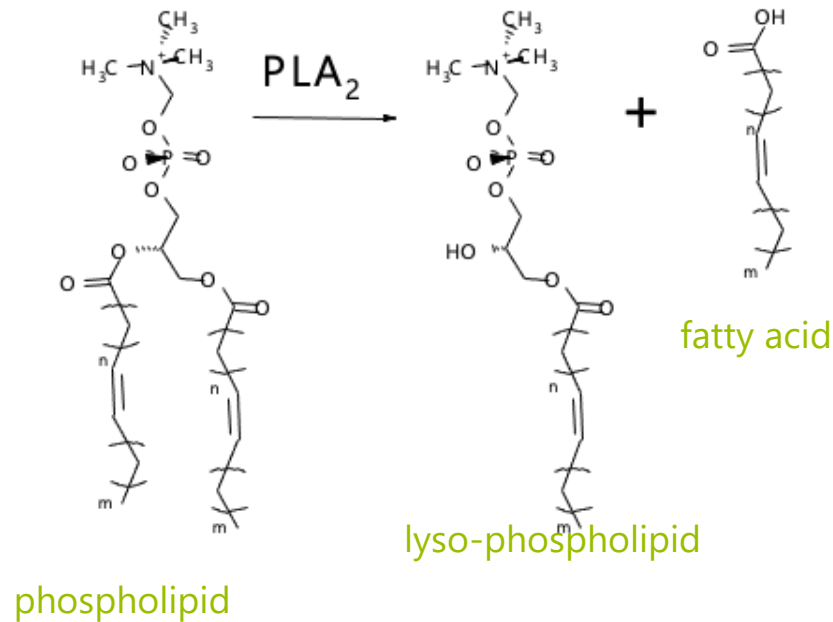
- 1: SiO₂
- 2: solvent layer
- 3: lipid heads
- 4: lipid chains+AmB
- 5: lipid heads
- 6: AmB/dErg



4 Examples 1: Phospholipase A₂

Understanding interfacial enzyme catalysis by selective phospholipid deuteration

PLA₂ breaks down phospholipids in digestion, inflammation, venoms, cell signalling etc.

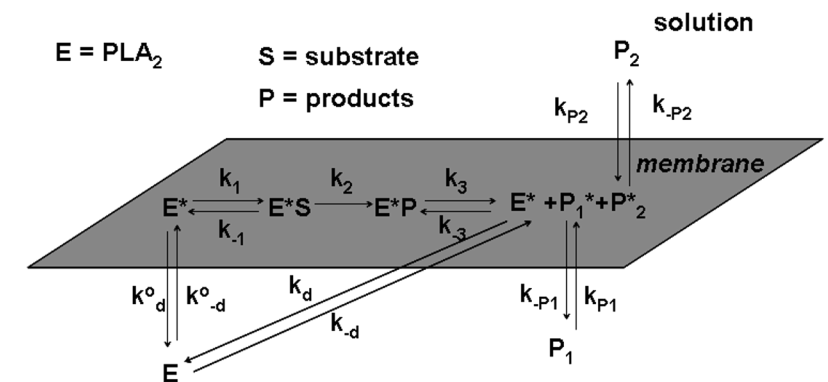


active site

Naja Naja Naja (Cobra)

Mw	13219
Vmol/ Å ³	15451

Interfacial enzyme catalysis:



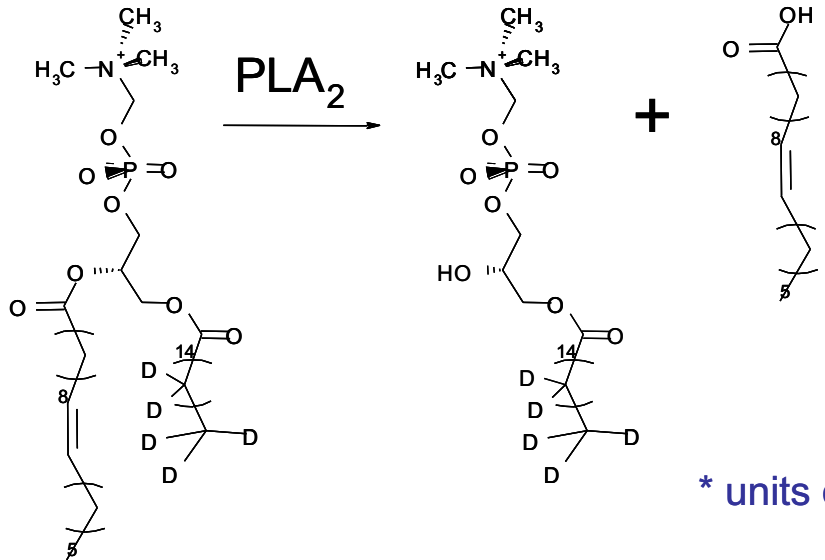
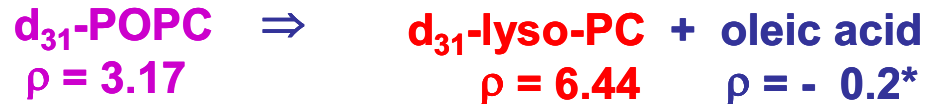
⇒ enzyme crystal structure is not enough to understand mechanism of PLA₂



4 Examples 1: Phospholipase A₂

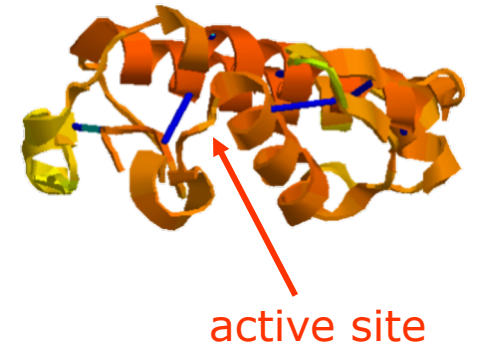
Understanding interfacial enzyme catalysis by selective phospholipid deuteration

PLA₂ breaks down phospholipids in digestion, inflammation, venoms, cell signalling etc.



* units of 10^6 \AA^{-2}

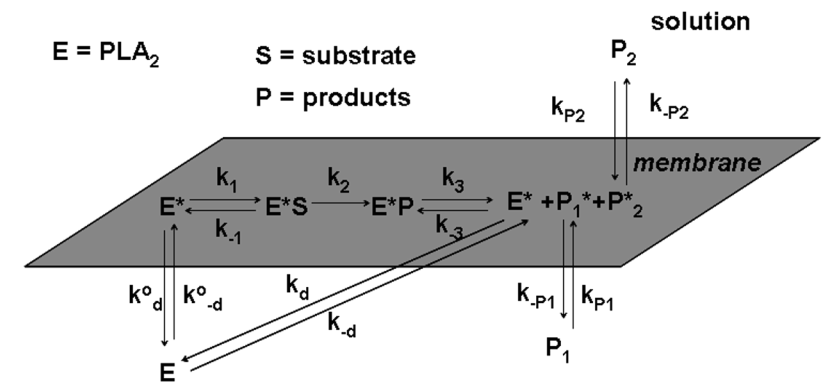
\Rightarrow neutron reflection to reveal what happens to reaction products



active site
Naja Naja Naja (Cobra)

Mw	13219
Vmol/ Å ³	15451

Interfacial enzyme catalysis:

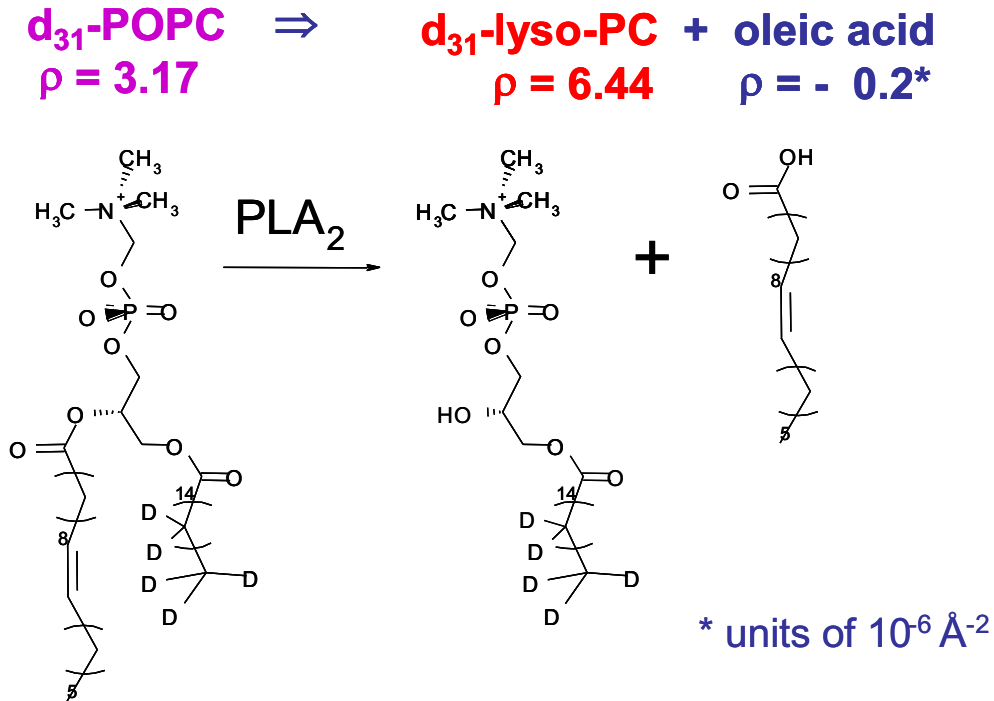


\Rightarrow enzyme crystal structure is not enough to understand mechanism of PLA₂

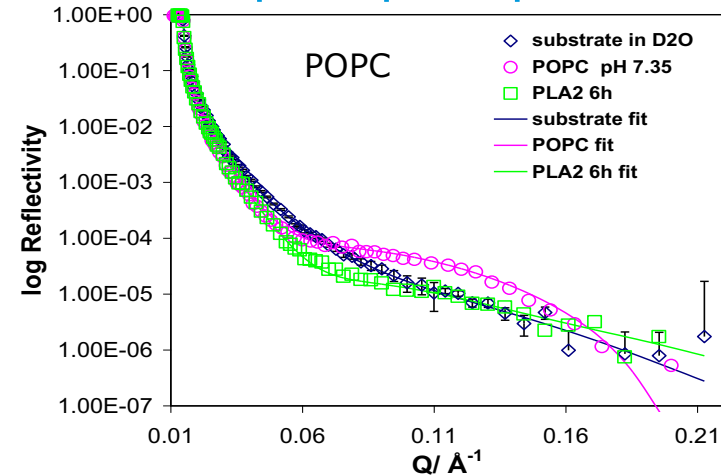
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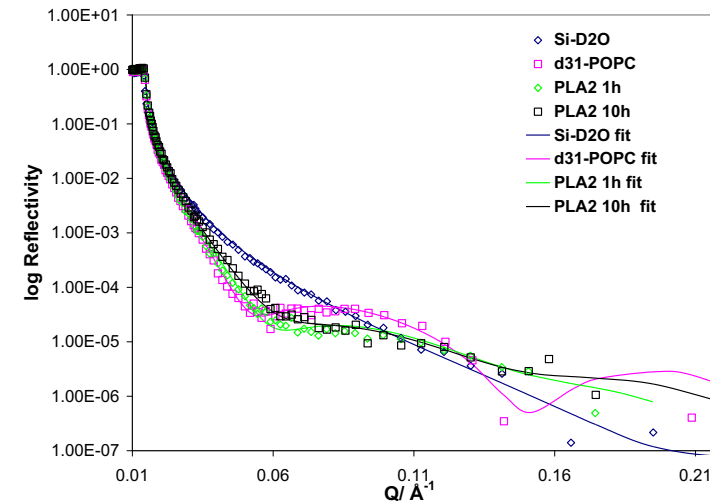


\Rightarrow neutron reflection to reveal what happens to reaction products



h-POPC in D₂O : reflectivity drops as ~50% of lipid membrane is destroyed

$d_{31}\text{-POPC}$ cobra PLA₂ pH 7

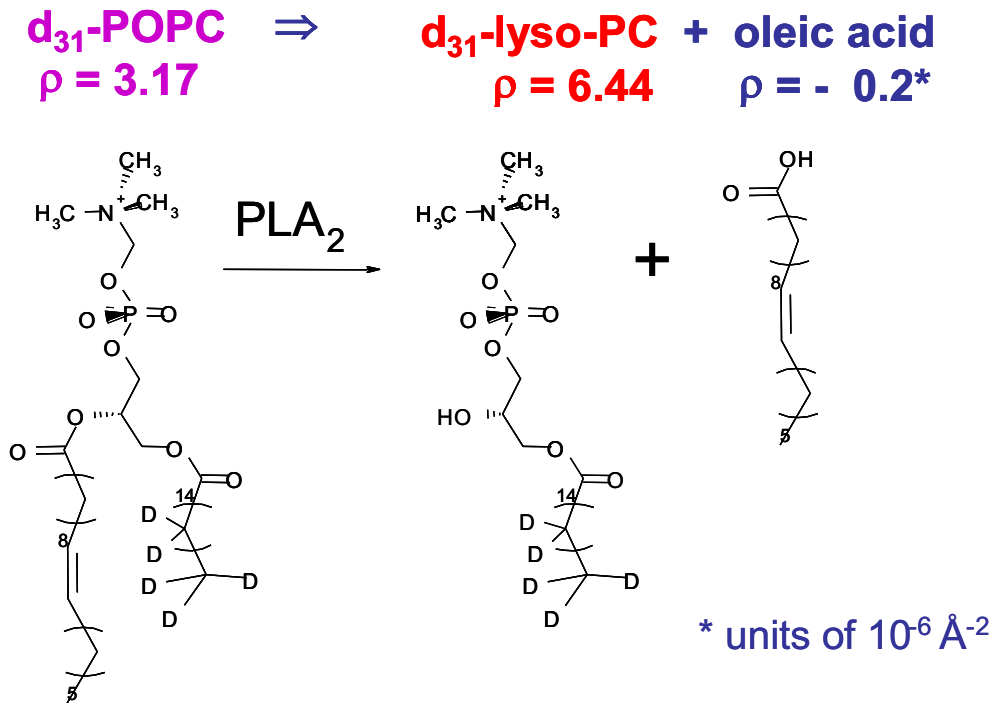


$d_{31}\text{-POPC}$ in D₂O: amount of lipid does not change much, but membrane gets thinner.

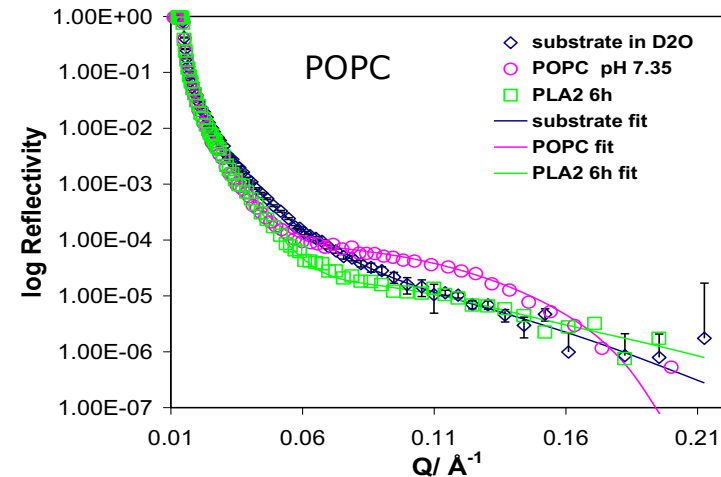
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Understanding interfacial enzyme catalysis by selective phospholipid deuteration

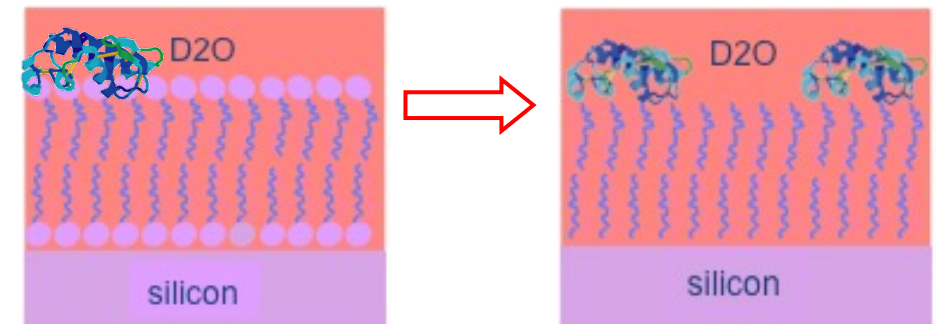
PLA₂ breaks down phospholipids in digestion, inflammation, venoms, cell signalling etc.



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h-POPC in D₂O : reflectivity drops as ~50% of lipid membrane is destroyed

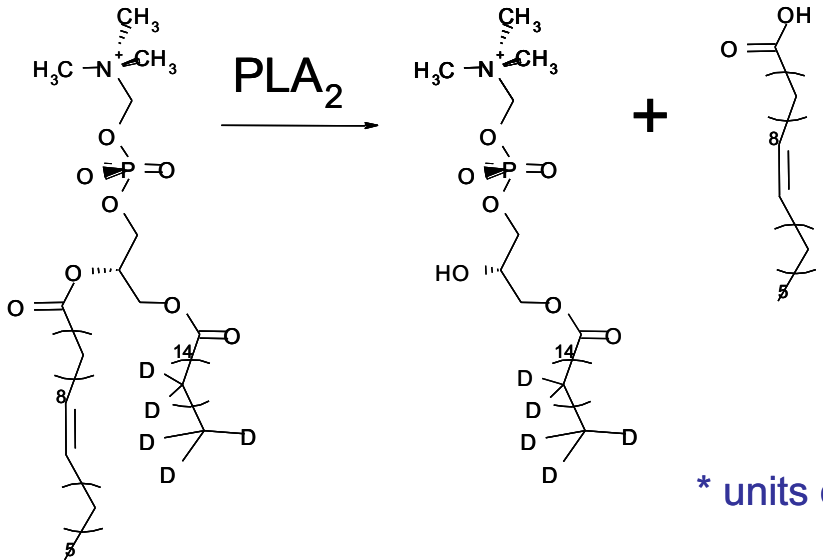
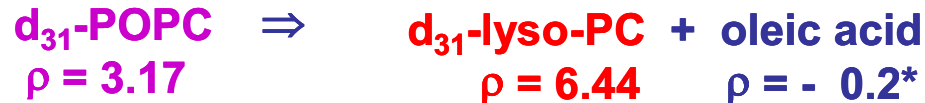


d₃₁-POPC in D₂O: amount of lipid does not change much, but membrane gets thinner.

4 Examples 1: Phospholipase A₂

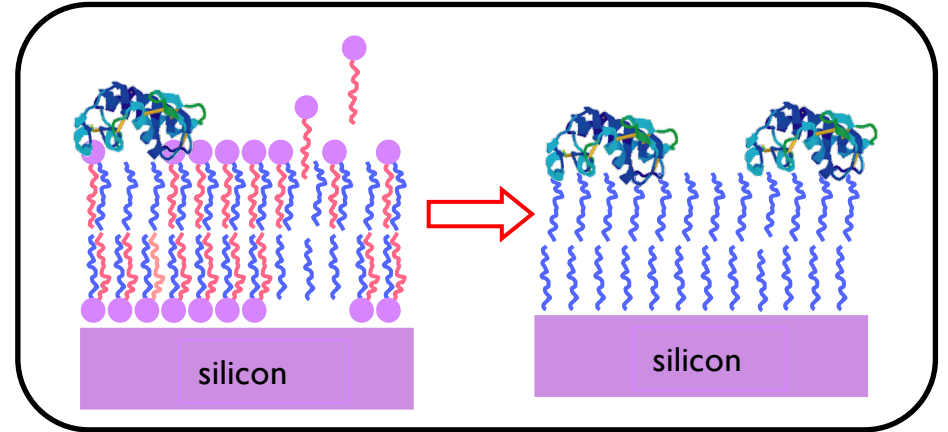
Understanding interfacial enzyme catalysis by selective phospholipid deuteration

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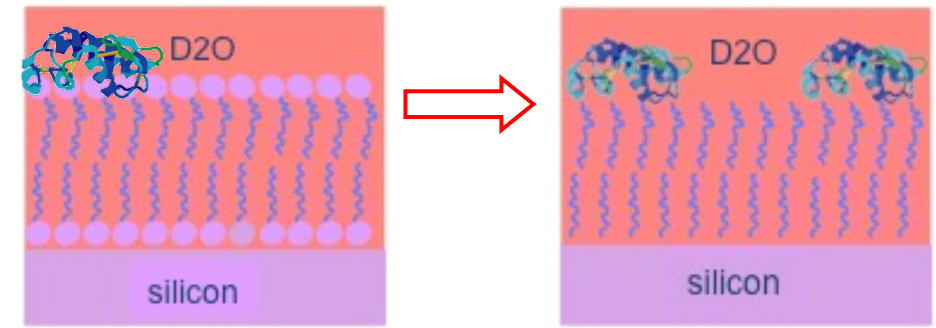


* units of 10^{-6} \AA^{-2}

\Rightarrow neutron reflection to reveal what happens to reaction products



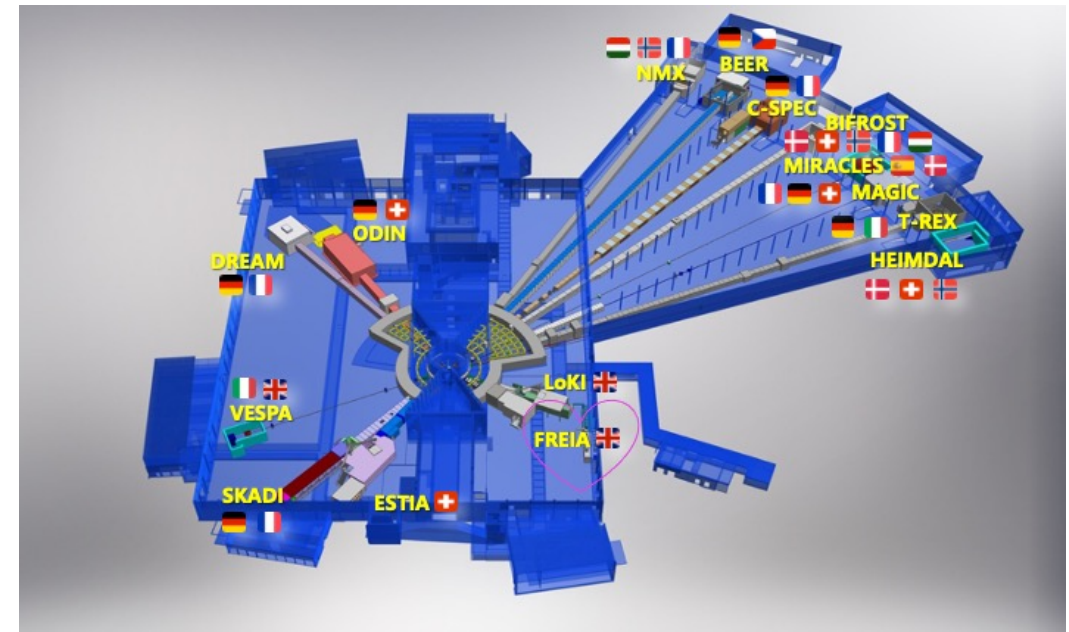
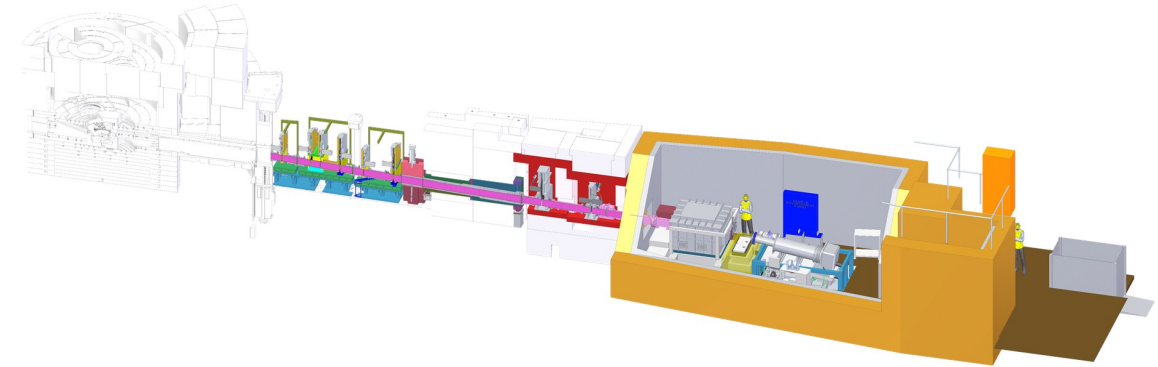
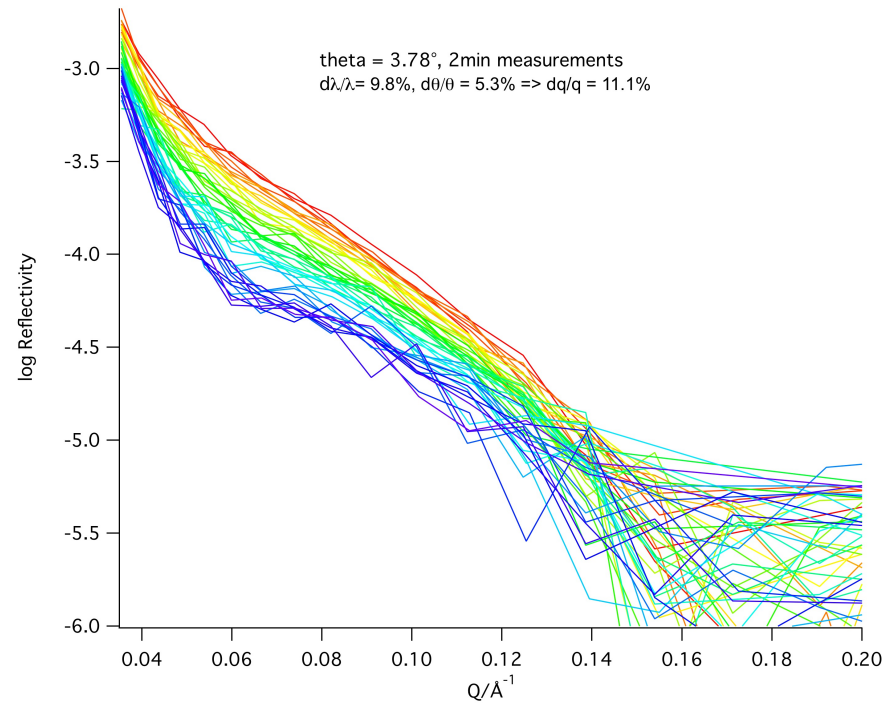
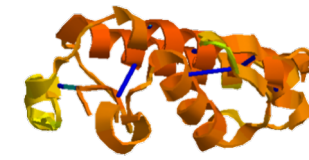
h-POPC in D₂O : reflectivity drops as ~50% of lipid membrane is destroyed



d₃₁-POPC in D₂O: amount of lipid does not change much, but membrane gets thinner.

Kinetics

location of enzyme and membrane structure as function of time?



- Measuring changing membrane structure and composition every 2 min (FIGARO @ ILL).
- On FREIA @ ESS we will be able to measure on sub-second timescales!

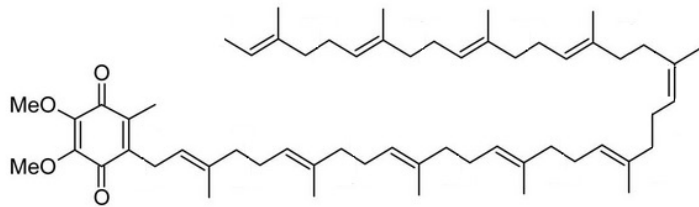
5 Examples 2: Membrane asymmetry

Resolving lipid membrane inner structure

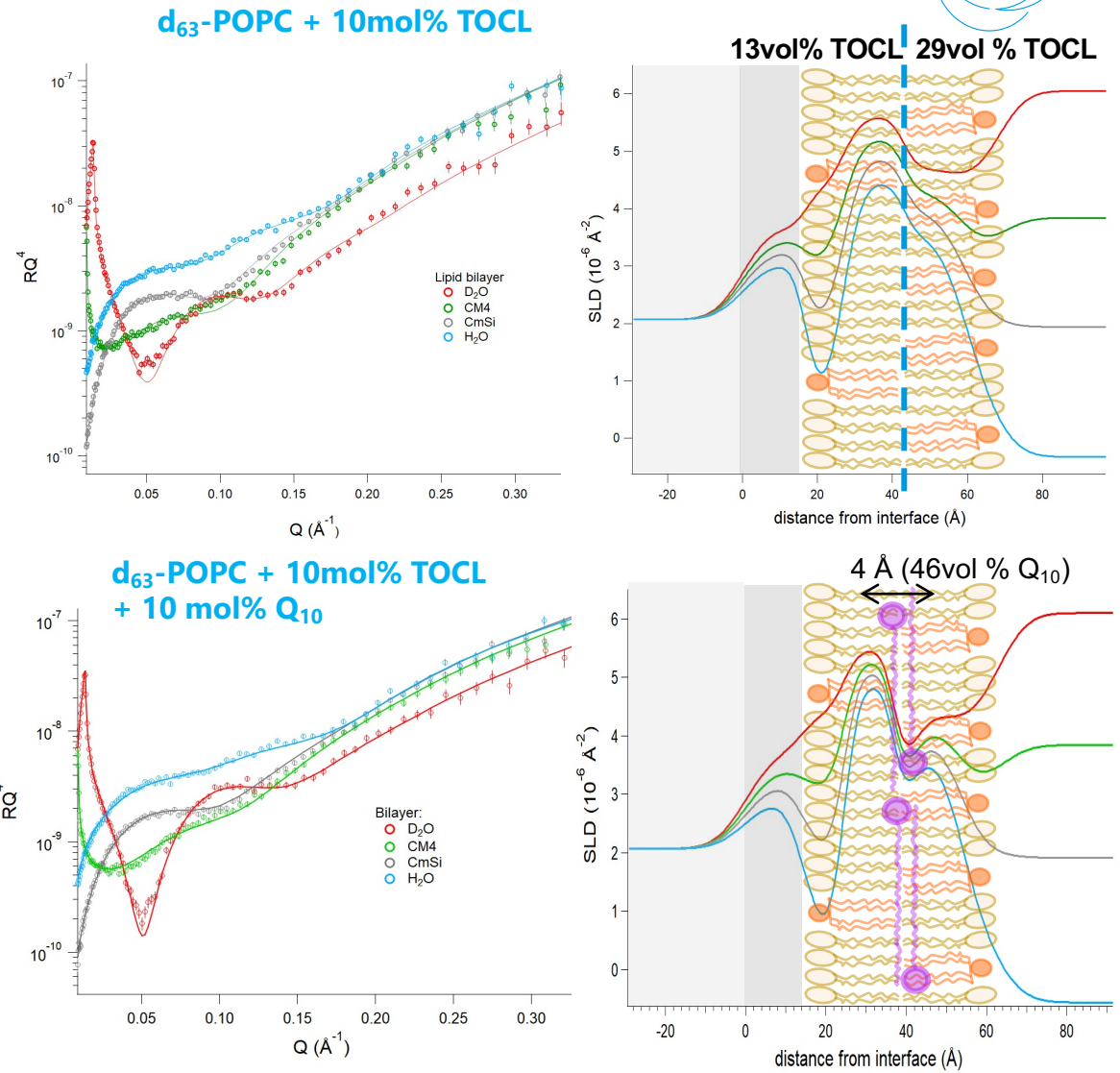
Lipid mixtures can self-assemble into asymmetric supported bilayers

- Cardiolipin is negatively charged and repelled by SiO₂ surfaces
- CL asymmetry detected using chain-deuterated POPC.

Ubiquinone Q₁₀ is a long hydrophobic electron acceptor for many enzymes in mitochondrial membranes.



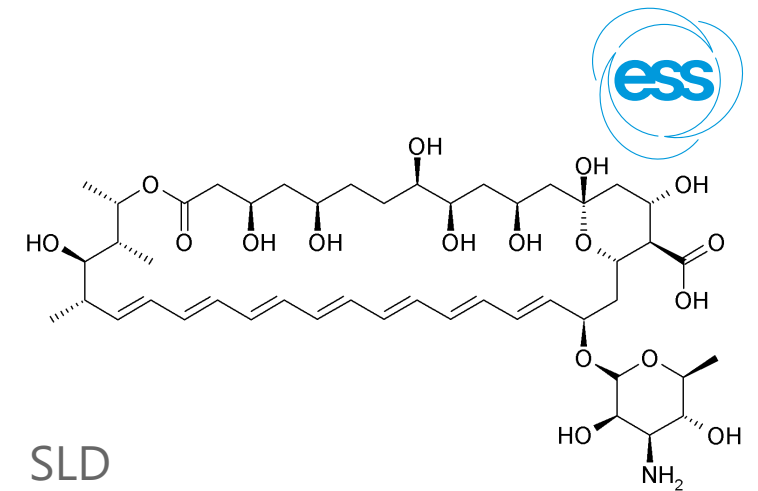
- Location in centre of membranes resolved by NR.



6 Example 3: Amphotericin B

Detecting antibiotic insertion and sterol extraction

- AmB is a membrane-binding antimyktotic drug that inserts into lipid bilayers
- AmB exchanges 13 of 47 protons with aqueous solvents
- sld varies in different solvent contrasts
- AmB has contrast to both hydrogenous and deuterated lipids in all contrasts
- AmB is contrast matched to silicon substrates and CmSi!
- AmB solvent exchange can be analysed to resolve membrane insertion

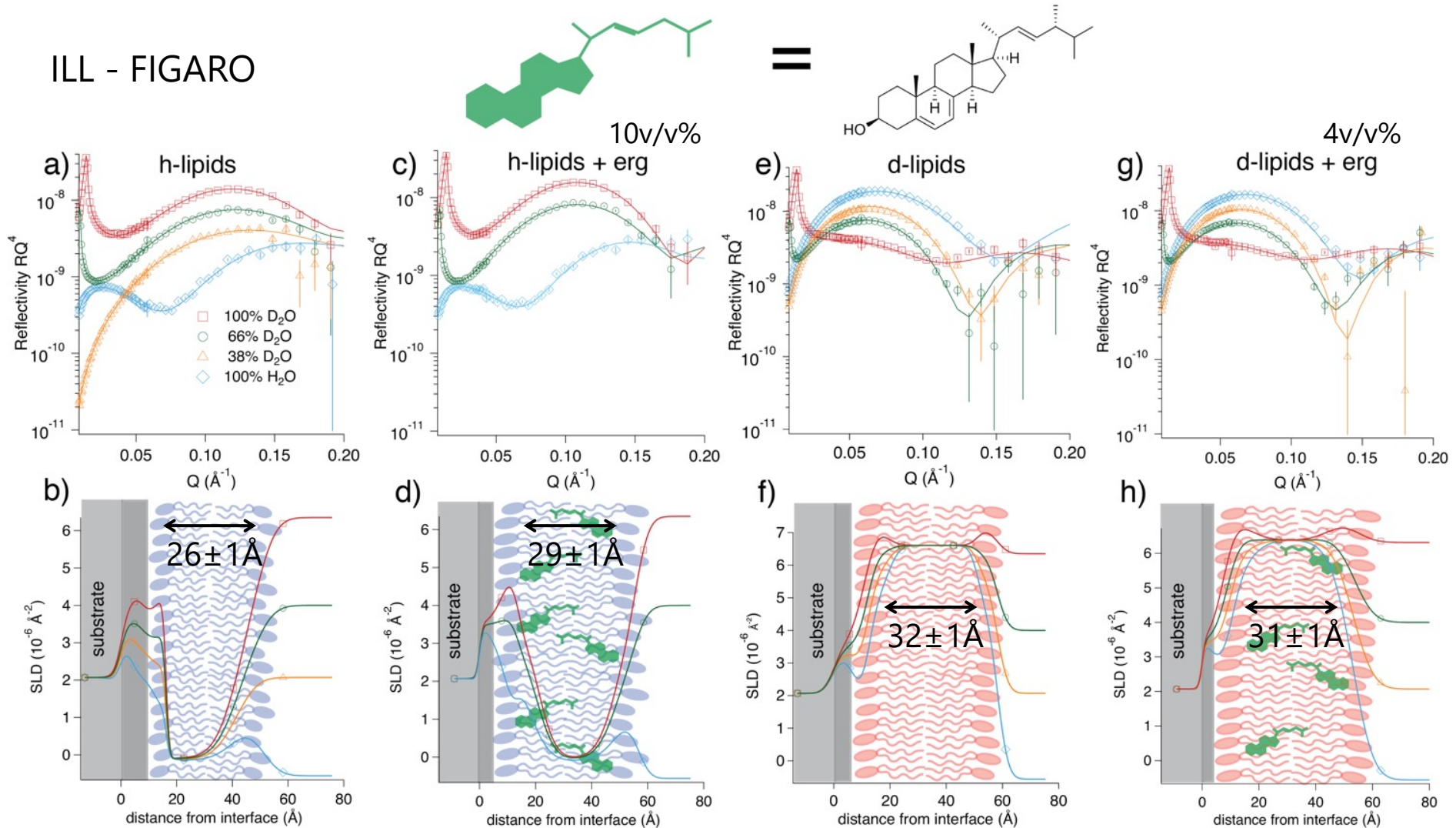


H ₂ O -0.56	CmSi (38% D ₂ O) 2.07	CM4 (66% D ₂ O) 4.0	D ₂ O 6.35
1.52	2.03	2.52	2.87

6 Example 3: Amphotericin B

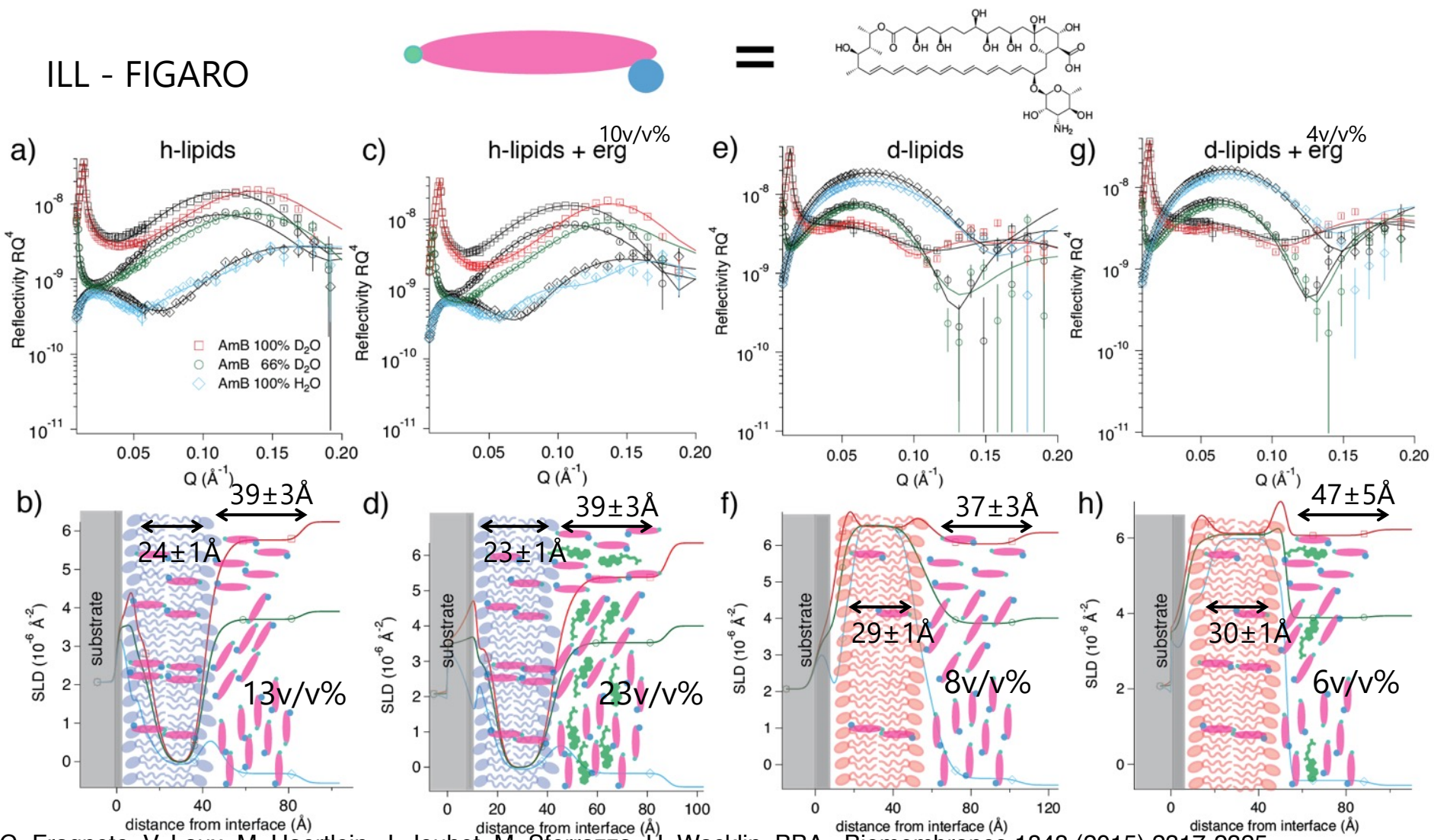


Structure of *P.Pastoris* yeast phospholipid membranes \pm native ergosterol



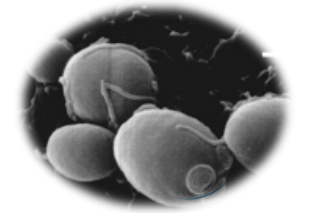
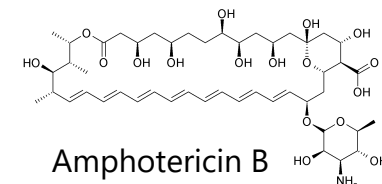
6 Example 3: Amphotericin B

AmB inserts into yeast lipid bilayers and extracts ergosterol into "sponge-layer"



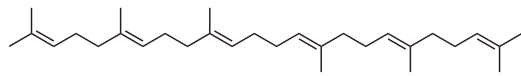
6 Example 3: Amphotericin B

Linking genes and lipid composition to Amphotericin B resistance in *C.glabrata*

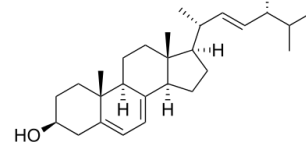


C. glabrata is a human pathogen attacking immunocompromised patients.

- *C.glabrata* used iRNA library of strains to up/down regulate genes
- Phospholipid/fatty acid composition shows minor changes
- AmB resistant *C. glabrata* accumulates squalene (1st precursor to all sterols)

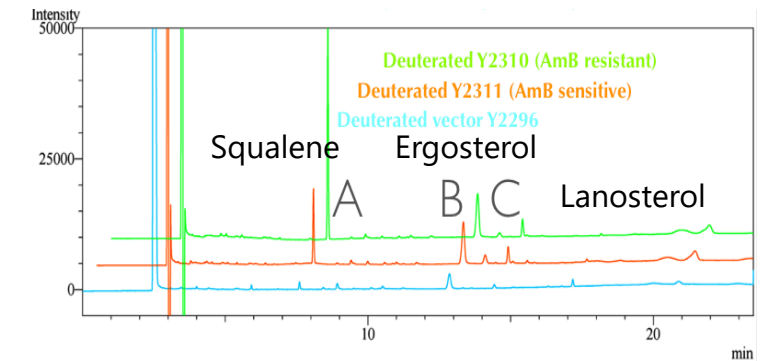
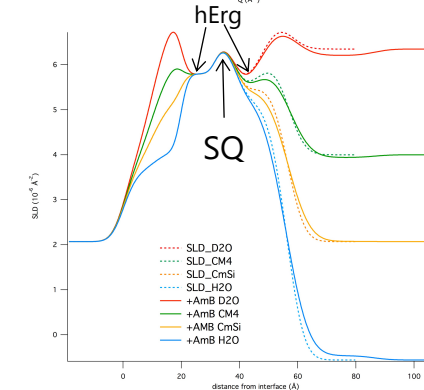
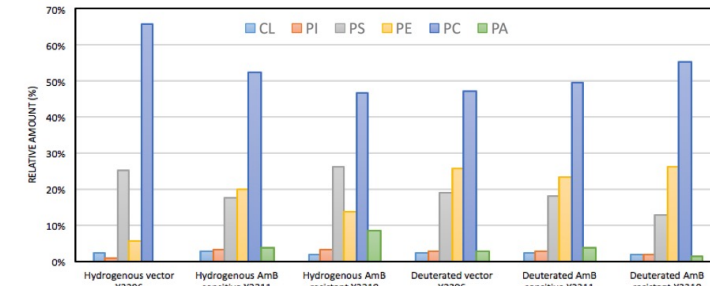
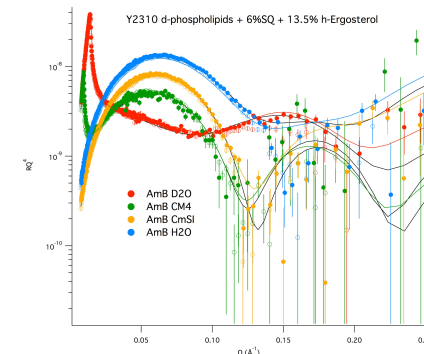
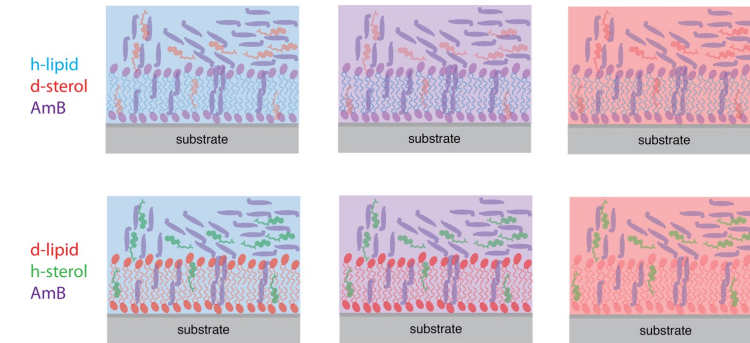
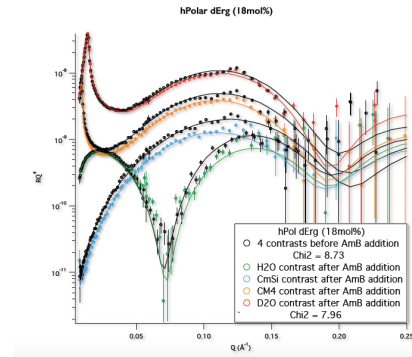


squalene



ergosterol

- NR: AmB extracts d₄₄ergosterol but this is partly blocked by squalene.
- *C.glabrata* resistance correlates with lower ergosterol extraction



7 Conclusion

Lipid deuteration is a powerful tool to understand membrane structure and function

Deuteration of a range of membrane lipids is possible synthetically, enzymatically and biologically.

- Synthetic lipids can be made in large quantities, but only few available and methods are time-consuming €€€
- Biological lipid extracts containing many lipid classes and a range of fatty acid chains – more native like models of cell membranes but purification of components is challenging €€€
- Contrast matched/selectively labelled lipids can be produced both synthetically and biologically
- *Analysis of deuteration, purity and molecular composition is important!*

Deuteration allows the high resolution of NR and SANS to be fully exploited:

- Detection of lipid arrangement in multicomponent membranes, e.g. asymmetry
- Changes in membrane lipid composition/arrangement
- Penetration of proteins/drugs etc.
- *Solvent contrast variation important for reducing ambiguity in analysis*
- Magnetic contrast variation for fragile samples where solvent exchange not possible, but long PNR measurements.
- *The analysis is as good as the information/assumptions used!*

5 What types of support is available at ESS?

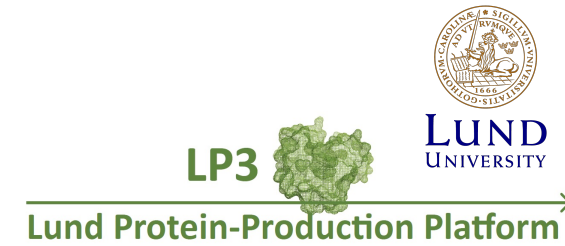
demax@ess.eu



Deuteration and crystallisation services are in operation since 2019:



ESS Deuteration and Macromolecular Crystallization (DEMAX) Platform

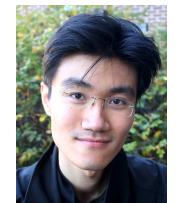


- ### Chemical Deuteration
- Catalytic H/D exchange
 - Organic (chemical) synthesis
 - Enzyme synthesis
 - Cell lipid extraction, separation and analysis
 - NMR, GC-MS (Red Glead, LU)

- ### Biological Deuteration
- Deuterated biomass (algae, bacteria, yeast)
 - Deuterated protein expression, purification
 - Biophysical characterization (DLS, nanoDSF, Thermoflour)
 - ²H-incorporation ESI-MS (KI)

- ### Macromolecular Crystallization
- High- and Low-throughput screening
 - Optimization (fine-screening, scale-up, temperature)
 - Testing & data collection with X-rays @ MAX IV lab (with LP3)

LP3 biotechnician



PD: Jia-Fei Poon



PD: Jenny Andersson



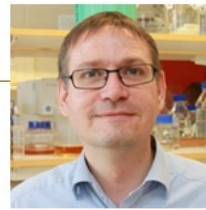
Anna Leung



Fatima Plieva



Hanna Wacklin-Knecht



Wolfgang Knecht



DEUNET international deuteration network



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Chemical & biological deuteration, crystal growth



Polymer & chem synthesis



Lipid biodeuteration (PSCM), biodeuteration (D-lab)



since 2019



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Chemical & biological
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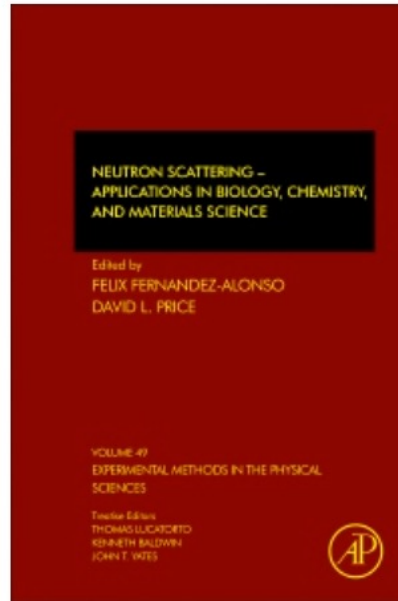
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LEAGUE OF ADVANCED
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Questions?